Total Synthesis of Bleomycin A_2 and Related Agents. 2. Synthesis of (-)-Pyrimidoblamic Acid, epi-(+)-Pyrimidoblamic Acid, (+)-Desacetamidopyrimidoblamic Acid, and (-)-Descarboxamidopyrimidoblamic Acid

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Abstract: Full details of concise syntheses of (-)-pyrimidoblamic acid (1), the authentic heterocyclic core of the bleomycin A_2 metal binding domain, as well as the key substructure analogs epi-(+)-pyrimidoblamic acid (2), (+)desacetamidopyrimidoblamic acid (3), and (-)-descarboxamidopyrimidoblamic acid (4) are described. Key to the approach is the implementation of an inverse electron demand [4+2] cycloaddition reaction of 2,4,6-tris(ethoxycarbonyl)-1,3,5-triazine with 1-(dibenzylamino)-1-propyne or in situ generated 1,1-diaminopropene for the one-step preparation of an appropriately functionalized pyrimidine nucleus. The development and subsequent implementation of a diastereoselective imine addition reaction of optically active N-acyloxazolidinone enolates provided a stereocontrolled introduction of the pyrimidoblamic acid C2 acetamido side chain. Chemical studies which unambiguously establish and confirm the absolute configuration of the C2 acetamido side chain are detailed, and their extension to the synthesis of (-)-descarboxamidopyrimidoblamic acid (4) is described.

The bleomycins are a family of glycopeptide antitumor antibiotics possessing clinically useful activity thought to be mediated through their metal-dependent oxidative cleavage of duplex DNA¹ (Figure 1). Consequently, bleomycin A_2 , its naturally occurring congeners, its semisynthetic derivatives and degradation products, and synthetic analogs have been the subject of extensive investigations in efforts to define the fundamental functional roles of its structural subunits. In the preceding article, we provided full details of concise syntheses of tri-, tetra-, and pentapeptide S as well as a series of structural analogs and the determination of their DNA binding properties.² Herein we provide full details of the synthesis of (-)-pyrimidoblamic acid (1),³⁻⁵ epi-(+)-pyrimidoblamic acid (2),^{4,5} (+)-desacetamidopyrimidoblamic acid (3),4,6 and (-)-descarboxamidopyrimidoblamic acid (4) in efforts that represent a systematic exploration of the key structural features of the metal binding domain of bleomycin A₂.¹ Extensive studies from the Umezawa-Ohno

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laboratories have defined a number of seminal features of the pyrimidoblamic subunit through chemical derivatization or degradation of the natural product⁷ and through chemical synthesis,⁸ and a number of additional simplified metal binding domains have been evaluated in the laboratories of others.4,5,9-13 Complementary to the efforts disclosed to date, our recent observation of the potentiating DNA cleavage effects of the C2

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acetamido side chain of P-3A¹³ and deglycobleomycin A_2^6 in addition to the proposed rapid metabolic inactivation of bleomycin A₂ through bleomycin hydrolase^{7b,14} hydrolysis of the carboxamide have suggested fundamental modifications of the pyrimidoblamic acid subunit that have yet to be explored. Concurrent with such efforts, a study of the bleomycin A2 analogs incorporating modified pyrimidoblamic acid subunits could be expected to permit an assessment of the role each substituent may play in contributing to metal chelation, O₂ activation, and DNA cleavage efficiency as well as the fundamental role the subunit may play in controlling the characteristic 5'-GC/GT DNA cleavage selectivity.¹⁵ The diastereoselective synthesis of (-)-pyrimidoblamic acid (1) detailed herein concisely installs the two remaining acyclic stereogenic centers of the natural aglycone. Notably, this represents the first reported approach which permits control of the relative and absolute stereochemistry of the C2 acetamido side chain of 1 and does so in a concise manner readily adaptable to the concurrent preparation of analogs. In realization of efforts to prepare by chemical synthesis bleomycin A₂ analogs possessing deep-seated structural changes for subsequent evaluation, the extension of the studies to the synthesis of 2-4 is additionally detailed. Pertinent to the studies detailed herein, the C2 acetamido side chain and the carboxamido group of the bleomycins have been shown not to be intimately involved in the key metal chelation and oxygen activation event required for DNA cleavage.¹ Thus, 2-4 constitute important substructures of the pyrimidoblamic acid subunit accessible only through chemical synthesis. These agents along with (-)-desmethylpyrimidoblamic acid, epi-(+)desmethylpyrimidoblamic acid, and (+)-desmethyldesacetamidopyrimidoblamic acid disclosed in recent efforts¹³ provide a key

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Scheme 1



series of structurally modified pyrimidoblamic acid analogs. Their incorporation into a full range of structural analogs of deglycobleomycin A_2 is detailed in the accompanying article.¹⁶

The approach to the authentic and modified pyrimidoblamic acid subunits is based on the inverse electron demand Diels-Alder reaction¹⁷ of 2,4,6-tris(ethoxycarbonyl)-1,3,5-triazine (5)^{13,18,19} for the preparation of the pyrimidine nucleus central to the structure of the agents. Key to the completion of the synthesis of (-)-pyrimidoblamic acid was the development and implementation of a diastereoselective imine addition reaction of optically active N-acyloxazolidinones for the stereocontrolled introduction of the C2 acetamido side chain.²⁰ Chemical studies which establish and confirm the absolute configuration of the C2 acetamido side chain are detailed, and their extension to the diastereoselective synthesis of 4 is described.

1,3,5-Triazine → Pyrimidine Heteroaromatic Diels-Alder Reaction: Synthesis of the Pyrimidine Core. Two concise approaches to the preparation of 9 based on the [4 + 2]cycloaddition of 2,4,6-tris(ethoxycarbonyl)-1,3,5-triazine (5) have been developed (Scheme 1). Treatment of 5, prepared in one step by the acid-catalyzed trimerization of ethyl cyanoformate (95-100%),¹⁸ with 1-(dibenzylamino)propyne (6,²¹ 2 equiv) provided 8 in excellent yield (95-98%) under thermal reaction conditions (101 °C, dioxane, 21 h). The room temperature [4 + 2] cycloaddition reaction of 5 with 6 is followed by a subsequent retro Diels-Alder reaction with loss of ethyl cyanoformate, and it is the rate of the cycloreversion reaction that dictates the required

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Table 1. Representative Results of a Study of the [4 + 2] Cycloaddition Reactions of 5 with 6 and 7

entry	conditions	product	yield (%)
1	DMF, 90 °C, 29 h	9	75
2	DMF, 100 °C, 72 h	9	80
3	DMF, 110 °C, 22 h	9	65
4	DMF, 120 °C, 48 h	9	53
5	DMF, 130 °C, 29 h	9	46
6	DMF, 140 °C, 26 h	9	44
7	DMF, K ₂ CO ₃ , 100 °C, 48 h	9	45
8	PhCH ₃ , CH ₃ CO ₂ H, 100 °C, 72 h	9	7
9	dioxane, 101 °C, 48 h	9	0
10	dioxane, 101 °C, 21 h	8	95
11	CH ₃ CN, 82 °C, 12 h	8	73

thermal reaction conditions. Acid-catalyzed debenzylation of 8 under vigorous conditions (CF₃SO₃H, CH₂Cl₂, 40 °C, 13 h, 75%) provided 9, and initial efforts to effect the conversion of 8 to 9 through catalytic hydrogenolysis proved less successful. In general, the deprotection of 8 under a range of alternative conditions gave predominantly the monodebenzylation product in addition to recovered starting material. Alternatively, 9 was derived directly and conveniently in one step by treatment of 5 with propionamidine hydrochloride (7, 100 °C, DMF, 72 h, 80%) in a reaction cascade that proceeds with thermal tautomerization of 7 to 1,1-diaminopropene and its [4 + 2] cycloaddition reaction with 5. The sequential elimination of ammonia, imine to enamine tautomerization, and subsequent retro Diels-Alder loss of ethyl cyanoformate under the reaction conditions provided 9 directly in excellent yield. The use of a polar aprotic solvent, the use of the hydrochloride salt of 7, and the carefully defined thermal conditions facilitate both the amidine tautomerization and aromatization of the initial cycloadduct (Table 1). Notably, the use of lower reaction temperatures (<80 °C) provided lower conversions to 9 presumably due to inadequate amidine to enamine tautomerization. Higher reaction temperatures (>100 °C) similarly provided lower conversions to 9 and presumably may be attributed to competitive reactions of the substrates, 9, and the reaction cascade intermediates under the more vigorous reaction conditions. The use of the propionamidine free base resulted in lower yields of 9 (45%, DMF, 100 °C, 48 h), and efforts to employ the corresponding methyl imidate or imidate hydrochloride proved significantly less successful.

Synthesis of (+)-Desacetamidopyrimidoblamic Acid (3). Key to the use of 9 in the synthesis of 1-4 was the selective differentiation of the pyrimidine C2 and C4 ethyl esters. This transformation was accomplished through selective reduction of the sterically and electronically more accessible C2 ethoxycarbonyl group of 9 to provide 10. Characteristic of the enhanced electrophilic nature of the C2 ethoxycarbonyl group responsible for the selective reduction, the reaction was effectively conducted with sodium borohydride at low temperature (1.0 equiv, EtOH, 5 °C, 150 h, 70%; Scheme 2). In a study of the reduction of 9, it was observed that the selectivity of the competitive C2 and C4 ethoxycarbonyl reductions improved as the reactivity of the reagent was reduced and the reaction temperature was lowered (Table 2). A number of more reactive reducing agents conventionally employed to reduce esters failed to provide the required C2 versus C4 reduction selectivity including LiBH₄, DIBAL·H, and L- and K-Selectride. 2D 1H-1H NOESY NMR of the isomeric alcohols unambiguously confirmed the isomer assignments through observation of a diagnostic CH₂OH/C5-CH₃ NOE crosspeak for the minor isomer and its absence with 10. Alternative efforts to differentiate the C2 and C4 ethyl esters through selective acid- or base-catalyzed hydrolysis, transesterification, or aminolysis proved less successful and less direct than the selective $NaBH_4$ reduction of 9.

Conversion of 10 to the tosylate 11 (99%), which proceeded cleanly without competitive sulfonamide generation, followed by Scheme 2



Table 2. Representative Results of the Selective Reduction of 9

entry	conditions ^a	2:4-CH ₂ OH	yield (%)
1	tBuOH-EtOH (5:1), 80 °C, 1 h	2:1	30
2	EtOH, 25 °C, 15 h	4:1	49
3	EtOH, 10 °C, 108 h	23:1	51
4	EtOH, 5 °C, 150 h	28:1	70
5	MeOH, -25 °C, 72 h	no reaction	0
6	Me4NBH4, EtOH, 25 °C, 96 h	17:1	36
7	Zn(BH ₄) ₂ , THF-Et ₂ O (1:1), 25 °C	6:1	10

^a NaBH₄ unless indicated otherwise.

clean displacement with 12¹³ (4 equiv, 2 equiv of NaHCO₃, CH₃CN, 25 °C, 10 h) and subsequent protection of the secondary amine provided 13 (91% overall for two steps). Hydrolysis of the ethyl ester (LiOH, 96–100%) provided the N^{α} , N^{β} -bis((*tert*butyloxy)carbonyl) derivative of desacetamidopyrimidoblamic acid (14), $[\alpha]^{22}_D - 9.2$ (c 0.25, CH₃OH) (lit^{4a} $[\alpha]^{25}_D - 8.9$ (c 1.25, CH₃OH)), and subsequent acid-catalyzed deprotection (3 N HCl-EtOAc, 25 °C, 30 min, 90%) of 14 provided (+)-desacetamidopyrimidoblamic acid (3), $[\alpha]^{22}_D + 8.1$ (c 0.15, 0.1N HCl).

Diastereoselective N-Acyloxazolidinone Enolate-Imine Addition Reaction: Preparation of (-)-Pyrimidoblamic Acid (1) and epi-(+)-Pyrimidoblamic Acid (2). The final strategic element necessary for completion of the synthesis of pyrimidoblamic acid was the stereocontrolled introduction of the C2 acetamido side chain. Prior studies have relied on nonselective chemical approaches requiring a separation of the resulting 1:1 mixture of diastereomers.⁴ In parallel with studies on the total synthesis of P-3A,¹³ we have examined the potential diastereoselective addition of optically active enolates with imines and found that the use of the Evans' N-acyloxazolidinones²² provided a diastereoselective imine addition reaction suitable for the C2 acetamido side chain introduction.²⁰

Oxidation of 10 (10 equiv of MnO_2 , CH_3CN , 82 °C, 3 h, 83%) followed by condensation of 15 with 12¹³ (98–100%) afforded 16 (Scheme 3). The modest conversions observed under standard MnO_2 oxidation conditions (10 equiv, CH_2Cl_2 , 25 °C, 25–40%) that may be attributed principally to the limited solubility of 10 and 15 were improved upon conducting the reaction in refluxing acetonitrile. Further use of dilute reaction conditions (0.05 M) eliminated a minor competitive self condensation reaction with imine formation and provided excellent conversions of 10 to 15. A range of alternative oxidants including PCC, PDC, Ba(MnO_4)₂ (CH_2Cl_2 , 25 °C, 25%), and TFAA–DMSO were not as successful at providing 15 due to the solubility properties of the substrate under conventional reaction conditions.

Addition of the stannous (Z)-enolate 17, generated by treatment

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Scheme 3



of the corresponding oxazolidinone²³ (1 equiv, THF, -20 °C, 1 h) with iPr_2NEt (2.2 equiv) in the presence of Sn(OTf)₂ (2.0 equiv), to 16 (0.5 equiv) provided a separable 87:13 mixture of the imine addition adducts 18a and 18b (THF, 0 °C, 12 h, 81-85%). The prescribed reaction conditions were derived only through considerable experimentation and ultimately provided the opportunity to use a minimal number of protecting groups for the potentially reactive functionality found in imine 16. Key to the success of the diastereoselective imine addition reaction^{24,25} was the use of the stannous enolate^{25c,26,27} (2.0 equiv) in the presence of two additional equivalents of Sn(OTf)₂, and under such conditions the major anti-18a adduct was found to slowly epimerize to syn-18a (0 °C, 12 h, 16:1 anti:syn-18a versus 0 °C,

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The diastereoselection observed in the imine addition reaction which we have investigated in some detail deserves further discussion. Unlike the boron enolate, which might be anticipated to react as a nonchelated enolate through a closed-chair transition state to provide the Evans' syn addition product, 22,23 the expanded coordination sphere of tin(II) like that of titanium(IV)28,29 was expected to permit reaction of the chelated enolate 17 with 16 with the additional complexation and activation of the imine to provide the non-Evans' syn imine addition product. However, the reaction provided not the chelated enolate syn imine addition product but rather the corresponding chelated enolate anti adduct as the major product. In retrospect, this potentially may be attributed to reaction of the imine in its preferred E configuration with both large imine substituents occupying axial positions in a closed-chair transition state further reinforced through additional internal chelation of the imine pyrimidinyl nitrogen with tin(II).²⁷ Although it is not possible to rule out reaction of the chelated (Z)-enolate 17 with imine 16 activated by external coordination to the added Lewis acid and proceeding through an open transition state,³⁰ the lack of reaction of the corresponding di-*n*-butylboronyl (Z)-enolate with 16 in the presence of added Lewis acid catalysts (Sn(OTf)₂, Et₂AlCl) suggests this may prove unlikely.

The assignment of the relative and absolute stereochemistry of the imine addition products was made based on the following observations. Reductive desulfurization of anti-18a and syn-18a both provided 19, which was found to possess the natural Sconfiguration at the newly introduced amine center through conversion to natural (-)-pyrimidoblamic acid (1). In addition, deliberate epimerization of the major addition product (anti-18a) provided syn-18a, isomeric at the thiomethyl center. In

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Scheme 4



contrast, desulfurization of anti-18b provided 22, which proved to be a diastereomer of 19. The relative stereochemistry of anti-18a, syn-18a, and anti-18b was established upon conversion to the cyclic carbamates 25-27 (Scheme 4).

For the cyclic carbamate 25 derived from the major addition product, the C4-H/C5-H coupling constant was determined to be 1.5 Hz, diagnostic of an equatorial-equatorial H-5/H-4 relationship, and proved analogous to that of 28 for which a singlecrystal X-ray structure had unambiguously established the trans stereochemistry.²⁰ Notably, the cyclic carbamate 25 adopts a conformation in which both the thiomethyl and aryl substituents occupy axial positions, and is the result of the lack of 1,3-diaxial MeS/H interactions and the presence of a single 1,3-diaxial Ar/H interaction within the preferred diaxial conformation. The alternative diequatorial conformation suffers from two significant and destabilizing gauche interactions of the vicinal ring substituents. Similarly, carbamate 27 derived from the minor imine addition product exhibits a C4-H/C5-H coupling constant of 1.2 Hz, characteristic of the trans stereochemistry, and, like 25, is derived from an imine addition product possessing the anti stereochemistry. In contrast, carbamate 26 derived from syn-18a exhibited a characteristic C4-H/C5-H coupling constant of 5.6 Hz, diagnostic of an axial-equatorial H-4/H-5 relationship in which H-5 occupies an axial position. Consistent with these assignments, the C5-H $w_{1/2}$ values of 6.8 Hz (25 and 27) and 19.4 Hz (26) were observed and proved diagnostic of an equatorial

and axial C5-H, respectively. Further consistent with the assignments, 25 and 27 exhibited ¹H-¹H NOE crosspeaks in the 2D-NOESY NMR for MeS/H-6a, H-5, and H-4. Notably, crosspeaks for MeS/H-6b and H-4/H-6a were not observed and would be diagnostic of an equatorial versus axial thiomethyl substituent and a 1,3-diaxial H-4/H-6a relationship, respectively. Carbamate 26 exhibited ¹H-¹H NOE crosspeaks in the 2D-NOESY NMR for MeS/H-6a, H-6b, H-4, and H-5 as well as a diagnostic H-6b/CO₂CH₂CH₃ crosspeak, and no H-4/H-6a crosspeak was observed. The former are consistent with an equatorial versus axial thiomethyl substituent, and the latter two observations are diagnostic of a conformation in which the aryl group occupies an axial position. Finally, calculated coupling constants for H-4/H-5, H-5/H-6a, and H-5/H-6b for the analogous MM2 low-energy conformations of 25 or 27 (obsd 1.5, 1.5, and 2.0 Hz; calcd 1.4, 2.6, and 1.4 Hz) and for 26 (obsd 5.6, 4.9, 8.2 Hz; calcd 4.3, 5.2, 11.7 Hz) agree with the observed coupling constants of the structural assignments.

Confirmation of the Absolute Stereochemistry. The absolute configuration assigned to the pyrimidoblamic acid and bleomycin A2 center which bears the C2 acetamido side chain was established in a single-crystal X-ray structure determination of 29^{31,32} derived from a chemical degradation product obtained under vigorous acidic conditions (6 N HCl, 105 °C, 20 h) resulting in partial (15-30%) epimerization of the amine center under consideration.33,34 This partial epimerization of the precursor to 29 coupled with the use of a nonheavy-atom derivative in the crystallographic solution of its absolute configuration has resulted in some concern as to the accuracy of the assigned absolute stereochemistry. This proved to be of special concern since additional spectroscopic properties of 29 provided a tentative assignment of the opposite absolute configuration.³¹ Unfortunately, the existing syntheses of 1 which provided a 1:1 mixture of isomers at the critical amine center have relied on correlation with the natural product to provide their stereochemical assignments and thus have not addressed the confirmation of the pyrimidoblamic acid assigned stereochemistry. In the course of our work, this uncertainty had become an independent concern in the ongoing NMR studies in which the preliminary assignment of the NOEs of bleomycin A2 metal complexes was not easily reconciled on the basis of assigned absolute stereochemistry.35 This long-standing ambiguity in the absolute stereochemistry of 1 as well as the inability of our own asymmetric synthesis of pyrimidoblamic acid to permit unambiguous assignment of absolute configuration at the C2 acetamido side chain center provided the incentive for us to establish it by an additional chemical means.

Chemical degradation of naturally derived bleomycin A2³⁶ under strong acidic conditions (6 N aqueous HCl, 105 °C, 20 h) followed by immediate esterification (3 N HCl, CH₃OH, reflux, 1 h) of the dicarboxylic acid 30, which itself proved difficult to isolate, provided 31, $[\alpha]^{25}_{365}$ +26 (c 0.15, 1 N HCl). In turn, acid-catalyzed hydrolysis of the readily purified diester 31 (3 N aqueous HCl, 100 °C, 1 h) provided 30, $[\alpha]^{25}_{365}$ +30 (c 0.2, 1 N HCl) (lit³¹ $[\alpha]^{20}_{365}$ +47.5 (c 0.2, 1 N HCl) (Scheme 5). Exposure of 20 to the same set of reaction conditions similarly provided 31 (75%), $[\alpha]^{25}_{365}$ +37 (c 0.15, 1 N HCl), $[\alpha]^{25}_{D}$ +3.4 (c 0.15, 1 N HCl), and **30** (94%), [α]²⁵₃₆₅ +40 (c 0.07, 1 N HCl), confirming the correlation of 18a, 19-21, and 1 with naturally derived material. In contrast, exposure of 23 to the same set of

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Scheme 5



Scheme 6



reaction conditions provided 32 (64%), $[\alpha]^{25}_{365}-39$ (c 0.14, 1 N HCl), $[\alpha]^{25}_{D}-3.7$ (c 0.14, 1 N HCl), and 33 (94%), $[\alpha]^{25}_{365}-40$ (c 0.08, 1 N HCl), enantiomeric with 31-30 and confirming the diastereomeric stereochemical assignments for 18b, 22-24, and 2.

With this set of authentic correlation samples in hand, we elected to prepare 30-31 and 32-33 by asymmetric synthesis to permit the unambiguous assignment of stereochemistry. Treatment of aldehyde 15 with the di-*n*-butylboronyl (Z)-enolate 34^{23} provided the Evans' (2S,3R)-syn-aldol adduct 35 as the only detectable product (59%, >20:1 syn) in which the resident chirality on the optically active oxazolidinone dictates the relative and absolute stereochemistry at the newly introduced centers with reaction of the nonchelated boron enolate through a closed-chair transition state (Scheme 6). Bu₃SnH reductive cleavage of the thiomethyl substituent followed by direct azide displacement³⁷ of the alcohol upon Mitsunobu activation³⁸ provided 37 with clean



inversion of the stereochemistry.³⁹ Removal of the optically active oxazolidinone through methanolysis followed by reduction of the azide to the corresponding amine provided diester **39**, $[\alpha]^{25}_{365}$ -54 (c 0.16, 1 N HCl), $[\alpha]^{25}_D$ -5.6 (c 0.16, 1 N HCl).⁴⁰ Acidcatalyzed hydrolysis of **39** (3 N aqueous HCl, 100 °C, 3 h) provided diacid **33**, $[\alpha]^{25}_{365}$ -59 (c 0.11, 1 N HCl), enantiomeric with naturally derived material. Thus, consistent with the original absolute stereochemical assignments, **30**, **31**, pyrimidoblamic acid, and the bleomycins possess the S configuration as shown. The use of the enolate **40**, enantiomeric with **34**, provided comparable intermediates possessing the natural S configuration as detailed in the following efforts.

Preparation of (-)-Descarboxamidopyrimidoblamic Acid (4) and Exploration of an Alternative Diastereospecific Synthesis of (-)-Pyrimidoblamic Acid (1). Given the enhanced diastereoselection achieved with use of the aldol versus imine addition reaction of the N-acyloxazolidinone enolates, we elected to investigate the potential of its implemention in an alternative synthesis of (-)-pyrimidoblamic acid (1). Further adaptation of this approach was anticipated to provide a synthesis of 4 through the divergent and late stage introduction of a modified C2 side chain not as easily accessible through use of our initial strategy. Treatment of 15 with the di-n-butylboronyl (Z)-enolate 40^{23} provided (2R,3S)-syn-41 as the only detectable reaction product (67%, >20:1) (Scheme 7). Reductive desulfurization of 41 effected by treatment with Bu₃SnH for removal of the thiomethyl group provided 42 (93%). Aminolysis of the N-acyloxazolidinone conducted at 0 °C (77%) followed by Mitsunobu activation of the alcohol 43 and azide displacement³⁷ with inversion of the stereochemistry provided 44 in excellent yield (84%) with no evidence of loss of stereochemical integrity at the reaction center and with observance of only a trace amount of the corresponding elimination product. The reverse order of reactions with Mitsunobu activation and azide displacement³⁷ of the alcohol 42

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⁽³⁹⁾ Some elimination (ca. 15-20%) accompanies the displacement reaction on 36, but no isomeric azide was detected. In the sequence leading to 4, this competitive reaction was reduced to a trace side reaction by conducting the displacement on the carboxamide 43.

⁽⁴⁰⁾ Efforts to convert the oxazolidinone to the methyl ester prior to azide introduction led to competitive elimination versus azide introduction, and efforts to reduce the azide prior to oxazolidinone methanolysis led to competitive hydrolysis of the oxazolidinone.

followed by aminolysis of the N-acyloxazolidinone also provided 44 but in lower conversions $(51 \times 40\%)$. Reduction of the azide to the corresponding amine 45 was accomplished best by mild catalytic hydrogenolysis (H2, Pd-C, CH3OH, 25°C, 1 h, 88%), but 45 was also obtained in good yield upon treatment of 44 with Ph₃P (H₂O-HOAc, THF, 25 °C, 16 h, 71%). Alkylation of 45 with N-BOC-2-bromoethylamine (K₂CO₃, DMF, 0.4 M, 25 °C, 48 h) provided 46. Alternative efforts to conduct the alkylation reaction with the corresponding tosylate (DMF, CH₃CN, EtOH, acetone; 2-4 days, 25 °C, 18-30%) provided lower conversions with recovered 45, and efforts to drive the reaction to completion using more vigorous reaction conditions led to competitive amine elimination (DMF, 50 °C) or dialkylation (DMF, 6-13 kbar, 25 °C). Hydrolysis (LiOH) of the ethyl ester followed by acidcatalyzed deprotection of 47, $[\alpha]^{25}$ -8.0 (c 0.2, H₂O), provided (-)-descarboxamidopyrimidoblamic acid (4), $[\alpha]^{25}D - 18$ (c 0.05, H₂O).

Efforts to extend this approach to an alternative synthesis of (-)-pyrimidoblamic acid itself have not yet proven successful (eq 1). Significant, although not exhaustive, efforts to alkylate 45

$$45 \qquad \underbrace{\begin{array}{c} \mathsf{NHBOC} \\ \mathsf{CONH}_2 \\ \mathsf{X} = \mathsf{Cl}, \mathsf{Br}, \mathsf{OTs} \end{array}}_{\mathsf{X} = \mathsf{Cl}, \mathsf{Br}, \mathsf{OTs}} \qquad \mathbf{20} \qquad (1)$$

with β -chloro-, β -bromo-, or β -tosyl-N-BOC-L-alanine amide under a wide variety of conditions provided little or no 20 and generally provided recovered 45 and products derived from decomposition (elimination) of the alkylating agent. Alternative efforts to activate the alcohol of 43 toward direct displacement with β -amino-N-BOC-L-alanine amide by formation of the corresponding tosylate (21%) suffered from competitive elimination.

The incorporation of 1-4 into the total synthesis of bleomycin A_2 and structurally related agents is detailed in the accompanying articles.

Experimental Section

2.4-Bis(ethoxycarbonyl)-6-(dibenzylamino)-5-methylpyrimidine (8). A solution of 2,4,6 tris(ethoxycarbonyl)-1,3,5-triazine¹⁸ (5, 4.11 mmol, 1.22g) in dioxane (10 mL) under Ar was treated with 1-(dibenzylamino). 1-propyne²¹ (6, 8.22 mmol, 1.93 g, 2.0 equiv) at 25 °C, and the resulting reaction solution was warmed at 101 °C (21 h). Removal of the solvent in vacuo and flash chromatography (SiO₂, 4×15 cm; 20-40% EtOAchexane gradient elution) afforded pure 8 (1.69 g, 1.78 g theoretical, 95%) as a colorless oil: ¹H NMR (CDCl₃, 300 MHz) & 7.32 (10H, m), 4.73 $(4H, s, 2 \times CH_2Ph), 4.45 (2H, q, J = 7.2 Hz, CH_2CH_3), 4.44 (2H, q, J)$ J = 7.2 Hz, CH_2CH_3), 2.36 (3H, s, CH₃), 1.42 (3H, t, J = 7.2 Hz, CH_2CH_3), 1.39 (3H, t, J = 7.2 Hz, CH_2CH_3); ¹³C NMR (CDCl₃, 75 MHz) & 165.8 (e, C-6), 165.7 (e, CO₂Et), 163.7 (e, CO₂Et), 157.9 (e, C-2), 153.2 (e, C-4), 136.9 (e), 128.5 (o), 127.9 (o) and 127.4 (o, four C_6H_5 , 116.3 (e, C-5), 62.1 (e, two CH_2CH_3), 52.4 (e, two CH_2Ph), 16.0 (o, CH₃), 14.1 (o, CH₂CH₃), 14.0 (o, CH₂CH₃); IR (film) *v*_{max} 3029, 2982, 2935, 1739, 1560, 1496, 1453, 1427, 1242, 1222, 1091, 800 cm⁻¹; UV (CHCl₃) λ_{max} 278 (ϵ 13 400), 244 (ϵ 6700) nm; CIMS (2methylpropane) m/e 434 (M+ + H, base); CIHRMS (2-methylpropane) m/e 434.2054 (M⁺ + H, C₂₅H₂₇N₃O₄ requires 434.2080).

Anal. Calcd for C₂₅H₂₇N₃O₄: C, 69.28; H, 6.24; N, 9.70. Found: C, 69.30; H, 6,29; N, 9.45.

6-Amino-2,4-bis(ethoxycarbonyl)-5-methylpyrimidine (9). Method A: A solution of 8 (1.42 mmol, 616 mg) in anhydrous CH_2Cl_2 (7 mL) was treated with CF_3SO_3H (14.2 mmol, 2.13 g, 10 equiv) at 25 °C, and the resulting reaction mixture was warmed at 40 °C (13 h). The mixture was cooled to 0 °C, diluted with H₂O (10 mL), and treated with 2 N aqueous NaOH (7.5 mL). Additional 2 N aqueous NaOH was added to further adjust the pH to 9, the mixture was extracted with CH_2Cl_2 (3 × 20 mL), and the combined organic phase was dried (MgSO₄). Removal of the solvent under reduced pressure and flash chromatography (SiO₂, 3 × 15 cm, 60% EtOAc-hexane) afforded pure 9 (270 mg, 359 mg theoretical, 75%) as a white solid identical to that described below.

Method B: A solution of 2,4,6 tris(ethoxycarbonyl) 1,3,5 triazine¹⁸ (5, 9.0 mmol, 2.69 g) in anhydrous DMF (45 mL) under Ar was treated with propionamidine hydrochloride (7, 18.0 mmol, 1.92 g, 2.0 equiv) at 25 °C, and the resulting reaction mixture was warmed at 100 °C (72 h). Removal of the solvent in vacuo and recrystallization (EtOAc-hexane) afforded pure 9 (1.82 g, 2.28 g theoretical, 80%) as a white solid: mp 155-156 °C (white needles, EtOAc-hexane); ¹H NMR (CDCl₃, 300 MHz) $\delta 6.22$ (2H, s, NH₂), 4.47 (2H, q, J = 7.1 Hz, CH₂CH₃), 4.45 (2H, q, J = 7.1 Hz, CH_2CH_3), 2.30 (3H, s, CH_3), 1.44 (3H, t, J = 7.1 Hz, CH_2CH_3), 1.43 (3H, t, J = 7.1 Hz, CH_2CH_3); ¹³C NMR (CDCl₃, 75 MHz) & 165.3 (e, C-6), 164.1 (e, C₄-CO₂Et), 163.7 (e, C₂·CO₂Et), 157.9 (e, C-2), 153.6 (e, C-4), 114.8 (e, C·5), 62.5 (e, CH₂CH₃), 62.2 (e, CH2CH3), 14.1 (0, CH2CH3), 14.0 (0, CH2CH3), 12.2 (0, CH3); UV (CHCl₃) λ_{max} 294 (ϵ 3900), 252 (ϵ 7100), 236 (ϵ 2700) nm; IR (KBr) vmax 3448, 3360, 2980, 2938, 1734, 1720, 1628, 1576 cm⁻¹; EIMS m/e (relative intensity) 253 (M⁺, 1), 181 (100), 135 (19), 107 (16), 81 (21); CIMS (2-methylpropane) m/e (relative intensity) 254 (M⁺ + H, base); EIHRMS m/e 253.1060 (M⁺, C₁₁H₁₅N₃O₄ requires 253.1062).

Anal. Calcd for $C_{11}H_{15}N_3O_4$: C, 52.12; H, 5.92; N, 16.58. Found: C, 51.91; H, 6.11; N, 16.34.

6-Amino-4-(ethoxycarbonyl).2-(hydroxymethyl).5-methyl. pyrimidine (10). A solution of 9 (2.0 mmol, 506 mg) in anhydrous EtOH (10 mL) was cooled to 5 °C and treated with NaBH₄ (2.0 mmol, 75.7 mg, 1.0 equiv) under Ar. After being stirred at 5 °C for 150 h, the reaction mixture was treated with saturated aqueous NaHCO₃ (10 mL) and aqueous H₂O₂ (5%, 10 mL) at 0 °C and was stirred for 5 h. The mixture was extracted with 20% 2-propanol-CHCl₃ (5 \times 20 mL), and the combined organic extracts were dried (MgSO₄). Removal of solvent in vacuo afforded a mixture of the 2- and 4-(hydroxymethyl)pyrimidines (2-CH₂OH:4-CH₂OH 28:1) as a white solid. The two isomeric alcohols were separated by preparative centrifugal thin-layer chromatography (SiO₂, 5% DMF-EtOAc) to afford pure 10 (312 mg, 422 mg theoretical, 70%) as a white solid: mp 169 °C sharp (white needles, EtOAc-hexane); ¹H NMR (CDCl₃, 300 MHz) & 5.77 (2H, br s, NH₂), 4.63 (2H, s, CH₂-OH), 4.43 (2H, q, J = 7.1 Hz, CH₂CH₃), 3.55 (1H, br s, OH), 2.22 (3H, s, CH₃), 1.41 (3H, t, J = 7.1 Hz, CH₂CH₃); 2D ¹H-¹H NOESY NMR (CDCl₃, 200 MHz) did not reveal an NOE crosspeak between CH₂OH and C5-CH₃ for 10 but did so for the minor isomer derived from the NaBH₄ reduction of 9; ¹³C NMR (DMSO- d_6 , 50 MHz) δ 166.9 (e, C=O), 166.5 (e, C-2), 164.1 (e, C-6), 154.2 (e, C-4), 108.7 (e, C-5), 64.6 (e, CH₂OH), 61.5 (e, CH₂CH₃), 14.2 (o, CH₂CH₃), 11.9 (o, CH₃); IR (KBr) vmax 3396, 3340, 3168, 2986, 2936, 1722, 1664, 1542, 1508 cm⁻¹; EIMS m/e (relative intensity) 211 (M⁺, 2), 139 (89), 110 (13), 108 (15), 85 (13), 81 (100); CIMS (2-methylpropane) m/e (relative intensity) 212 $(M^+ + H, base)$; EIHRMS m/e 211.0956 $(M^+, C_9H_{13}N_3O_3 requires)$ 211.0956).

Anal. Calcd for C₉H₁₃N₃O₃: C, 51.18; H, 6.16; N, 19.91. Found: C, 50.93, H, 6.39; N, 19.62.

6-Amino-4-(ethoxycarbonyl)-5-methyl-2-(((p-tolylsulfonyl)oxy)methyl)pyrimidine (11). A solution of 10 (2.56 mmol, 540 mg) in CH₂Cl₂ (6 mL) was treated with K₂CO₃ (5.12 mmol, 707 mg, 2.0 equiv) and p-TsCl (2.56 mmol, 488 mg, 1.0 equiv) at 25 °C under Ar, and the resulting reaction mixture was stirred at 25 °C for 20 h. The crude reaction mixture was filtered through a Celite pad (CH₂Cl₂, 3×10 mL), and the solvent was removed in vacuo. Chromatography (SiO₂, 3×7 cm, 50% EtOAc-hexane) afforded pure 11 (1.00 g, 1.01 g theoretical, 99%) as a white solid: mp 128 °C sharp (white needles, EtOAc-hexane); ¹H NMR $(CDCl_3, 300 \text{ MHz}) \delta 7.84 (2H, d, J = 8.1 \text{ Hz}, SO_2C_6H_4CH_3), 7.33 (2H, d)$ d, J = 8.3 Hz, $SO_2C_6H_4CH_3$), 6.31 (2H, br s, NH₂), 5.01 (2H, s, CH₂-OTs), 4.41 (2H, q, J = 7.2 Hz, CH_2CH_3), 2.44 (3H, s, $SO_2C_6H_4CH_3$), 2.22 (3H, s, CH₃), 1.40 (3H, t, J = 7.2 Hz, CH₂CH₃); ¹³C NMR (CDCl₃, 100 MHz) δ 165.7 (e, C==O), 164.4 (e, C-2), 159.1 (e, C-6), 152.8 (e, C-4), 145.5 (e), 131.6 (e), 130.0 (o), 128.1 (o), 112.0 (e, C-5), 70.7 (e, CH₂), 62.0 (e, CH₂CH₃), 21.6 (o, SO₂C₆H₄CH₃), 14.1 (o, CH₂CH₃), 11.7 (o, CH₃); IR (KBr) v_{max} 3457, 3328, 3181, 2980, 1720, 1654, 1570, 1559 cm⁻¹; CIMS (2-methylpropane) m/e (relative intensity) 366 (M⁺ + H, 18), 230 (base); FABHRMS (NBA) m/e 366.1124 (M⁺ + H, C₁₆H₁₉N₃O₅S requires 366.1124).

Anal. Calcd for $C_{16}H_{19}N_3O_5S$: C, 52.60; H, 5.21; N, 11.51; S, 8.77. Found: C, 52.84; H, 5.37; N, 11.14; S, 8.69.

 N^{α} , N^{β} -Bis((*tert*-butyloxy)carbonyl)- N^{β} -[(6-amino-4-(ethoxycarbonyl)-5-methylpyrimidin-2-yl)methyl]-(5)- β -aminoalanine Amide (13). A solution of 11 (0.63 mmol, 248 mg) and 12¹³ (2.5 mmol, 508 mg, 4.0 equiv) in CH₃CN (5.0 mL) was treated with NaHCO₃ (1.25 mmol, 105 mg, 2 equiv), and the mixture was stirred under Ar at 25 °C for 10 h. Removal of solvent in vacuo afforded the crude product, which was dissolved in

THF-saturated aqueous NaHCO3 (1:1, 5 mL) and treated with di-tertbutyl dicarbonate (2.5 mmol, 550 mg, 0.57 mL, 4 equiv), and the resulting mixture was stirred at 25 °C for 8 h. Water (5 mL) was added, and the mixture was extracted with 20% 2-propanol-CHCl₃ (3×20 mL). The combined organic extracts were dried (MgSO₄), and the solvent was removed in vacuo. Chromatography (SiO₂, 1 × 3 cm, 50-100% EtOAchexane gradient elution) afforded 13⁴¹ (276 mg, 310 mg theoretical, 89%; typically 89-91% for two steps) as a white solid: mp 174 °C sharp (EtOAc-hexane); $[\alpha]^{22}_{D}$ +27.9 (c 0.19, CHCl₃); ¹H NMR (DMSO-d₆, 200 MHz) & 7.50-6.70 (5H, m, NH2, CONH2, NHBOC), 4.40-4.10 $(3H, m, CH_2, CH_2CH), 4.30 (2H, q, J = 7.2 Hz, CH_2CH_3), 3.62 (1H, d)$ dd, J = 5.7, 13.9 Hz, CHHCH), 3.45-3.25 (1H, m, CHHCH), 2.00 (3H, s, CH₃), 1.45–1.20 (21H, m, CH₂CH₃, 2 × C(CH₃)₃); ¹H NMR (acetoned₆, 200 MHz) § 7.80-6.40 (5H, m), 4.80-4.10 (3H, m), 4.35 (2H, q, J = 7.0 Hz, 3.90 (1H, dd, J = 6.0, 14.0 Hz), 3.60–3.40 (1H, m), 2.15 (3H, s), 1.45–1.20 (21H, m); ¹³C NMR (DMSO- d_6 , 50 MHz) δ 172.9 (e, C=O), 166.7 (e, C=O), 164.7 (e, C=O), 164.3 (e, C=O), 164.0 (e, C-2), 155.8 (e, C·6), 154.1 (e, C·4), 108.4 (o, C-5), 79.4 (e, C(CH₃)₃), 79.0 (e, C(CH₃)₃), 78.5 (e, CH₂), 78.4 (e, CH₂CH), 61.4 (e, CH₂CH₃), 53.3 (o, CH₂CH), 28.3 (o, CH₃), 28.0 (o, CH₂CH₃), 14.0 (o, C(CH₃)₃), 11.8 (o, C(CH₃)₃); IR (KBr) ν_{max} 3358, 2978, 2934, 1686, 1650, 1582 cm⁻¹; CIMS (2·methylpropane) m/e (relative intensity) 497 (M⁺ + H, base); CIHRMS (2-methylpropane) m/e 497.2714 (M⁺ + H, C₂₂H₃₆N₆O₇ requires 497.2724).

Anal. Calcd for $C_{22}H_{36}N_6O_7:$ C, 53.23; H, 7.26; N, 16.94. Found: C, 53.07; H, 7.30; N, 16.71.

 N^{α}, N^{β} -Bis((*tert*-butyloxy)carbonyl)- N^{β} -[(6-amino-4-carboxypyrimidin-2-yl)methyl]-(S)- β -aminoalanine Amide (14). A solution of 13 (0.093) mmol, 46 mg) in THF-CH₃OH-H₂O (3:1:1, 0.5 mL) at 25°C was treated with 1 N aqueous LiOH (0.19 mmol, 0.19 mL, 2 equiv), and the mixture was stirred for 4 h. After evaporation of most of the THF-CH₃OH, the aqueous phase was extracted with CHCl₃ (2×2 mL). The aqueous phase was acidified to pH 4-5 with the addition of 1.2 N aqueous HCl and was extracted with 20% 2-propanol-CHCl₃ (5 \times 10 mL). The combined organic extracts were dried (MgSO₄), and the solvent was removed in vacuo to afford 14 (42 mg, 44 mg theoretical, 96%) as a white foam: $[\alpha]^{22}_{D}$ -9.2 (c 0.25, CH₃OH) (lit⁴ $[\alpha]^{25}_{D}$ -8.9 (c 1.25, CH₃OH)); Rf0.18 (SiO₂, 7:3 CHCl₃-CH₃OH); ¹H NMR (D₂O, 400 MHz) δ 4.33-4.39 (3H, m, CH2NBOC, CH2CH), 3.5-3.7 (2H, m, CH2CH), 2.09 (3H, br s, CH₃), 1.27–1.13 (18H, m, 2 × C(CH₃)₃); IR (CH₃OH) ν_{max} 3506, 3412, 3130, 2978, 2896, 1695, 1684, 1519, 1472, 1425, 1360, 1249, 1161, 1043, 732, cm⁻¹; FABHRMS (NBA) m/e 469.2420 (M⁺ + H, C₂₀H₃₂N₆O₇ requires 469.2411).

(+)-Desacetamidopyrimidoblamic Acid (3). A solution of 14 (0.021 mmol, 9.9 mg) in EtOAc (0.1 mL) was cooled to 0 °C and treated with 3 N HCl in EtOAc (3 mL). The resulting reaction mixture was stirred at 25 °C for 30 min. Removal of solvent in vacuo afforded 3 as a white solid. Pure 3 (7.1 mg, 7.9 mg theoretical, 90%) was obtained by trituration with anhydrous CH₂Cl₂ (3 × 5 mL) and Et₂O (3 × 5 mL): $[\alpha]^{22}_{D}$ +1.8 (c 0.55, CH₃OH), $[\alpha]^{22}_{D}$ +8.1 (c 0.15, 0.1 N HCl); ¹H NMR (CD₃OD, 400 MHz) δ 4.47 (1H, m, CH₂CH), 4.41 (2H, br s, CH₂), 3.65 (2H, m, CH₂CH), 2.41 (3H, br s, CH₃); IR (KBr) ν_{max} 3628, 3417, 3347, 3006, 1716, 1690, 1684, 1652, 1636, 1576, 1506, 1457, 1395, 1109, 1090 cm⁻¹; FABHRMS (NBA–CsI) *m/e* 401.0338 (M⁺ + Cs, C₁₀H₁₆N₆O₃ requires 401.0342).

6-Amino-4 (ethoxycarbonyi)-5-methylpyrimidine-2-carboxaldehyde (15). A solution of 10 (0.3 mmol, 71 mg) in anhydrous CH₃CN (6.7 mL) was treated with activated MnO₂ (3.36 mmol, 293 mg, 10 equiv) at 25 °C, and the resulting suspension was warmed at 82 °C for 3 h. The cooled reaction mixture was filtered through a Celite pad (CH₃CN, 5 × 10 mL), and the solvent was removed in vacuo. Flash chromatography (SiO₂, 0.5 × 3 cm, 80% EtOAc-hexane) afforded pure 15 (58 mg, 70 mg theoretical, 83%) as a white foam: ¹H NMR (CDCl₃, 300 MHz) δ 9.93 (1H, s, CHO), 5.75 (2H, br s, NH₂), 4.49 (2H, q, J = 7.1 Hz, OCH₂CH₃), 2.33 (3H, s, CH₃), 1.45 (3H, t, J = 7.1 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, 50 MHz) δ 192.2 (o, CHO), 165.6 (e, CO₂Et), 164.5 (e, c-2), 157.0 (e, C-6), 153.9 (e, c-4), 116.2 (e, c-5), 62.4 (e, CH₂CH₃), 29.5 (o, CH₃), 13.9 (o, CH₂CH₃); IR (KBr) ν_{max} 351, 3009, 2973, 2932, 2902, 1731, 1696, 1656, 1621 cm⁻¹; EIMS *m/e* (relative intensity) 209 (M⁺, 4), 180 (5), 137 (57), 109 (16), 92 (20), 81 (82), 69 (35), 57 (base); CIMS (2-methylpropane) m/e (relative intensity) 210 (M⁺ + H, base); EIHRMS m/e 209.0800 (M⁺, C₉H₁₁N₃O₃ requires 209.0800).

Anal. Calcd for $C_9H_{11}N_3O_3$: C, 51.65; H, 5.30; N, 20.10. Found: C, 51.60; H, 5.48; N, 20.05.

N^x-((tert-Butyloxy)carbonyl)-N^g.[[(6-amino-4-(ethoxycarbonyl)-5methylpyrimidin 2-yl)methylenejaminoj-(S)-β-aminoalanine Amide (16). A solution of 15 (0.215 mmol, 45 mg) and 1213 (0.215 mmol, 44 mg, 1.0 equiv) in anhydrous CH₃CN (4 mL) was treated with 4-Å molecular sieves (500 mg) at 25 °C. The resulting suspension was stirred at 25 °C for 24 h. The reaction mixture was filtered through a Celite pad (CH2-Cl₂, 5×5 mL). Removal of solvent in vacuo afforded 16 (83 mg, 84.7 mg theoretical, 98%) as a white foam: $[\alpha]^{25}D + 92.1$ (c 0.45, CHCl₃); ¹H NMR (CDCl₃, 400 MHz) δ 8.28 (1H, s, CH=N), 7.06 (2H, m), 5.71 (1H, d, J = 8.0 Hz, HNBOC), 4.58 (1H, m, CH₂CH), 4.45 (2H, q, J)= 7.1 Hz, CH_2CH_3), 4.11 (1H, dd, J = 3.3, 13.1 Hz, CHHCH), 3.93 (1H, dd, J = 5.6, 13.1 Hz, CHHCH), 2.28 (3H, s, CH₃), 1.41 (12H, m, CH₃, C(CH₃)₃); ¹³C NMR (CDCl₃, 75 MHz) δ 174.1 (e, C=N), 173.4 (e, C=O), 165.8 (e, C=O), 164.2 (e, C-2), 164.0 (e, C-6), 158.0 (e, C=O), 152.8 (e, C-4), 114.9 (e, C-5), 80.0 (e, C(CH₃)₃), 62.4 (e, CH₂-CH₃), 62.0 (o, CH₂CH), 55.3 (e, CH₂CH), 28.2 (o, CH₃), 28.1 (o, $C(CH_3)_3$, 14.0 (o, CH_2CH_3); IR (CHCl₃) ν_{max} 3619, 3415, 3010, 2978, 1725, 1687, 1577, 1501, 1422, 1215, 1072, 1047, 928 cm⁻¹; FABHRMS (NBA-CsI) m/e 527.1019 (M⁺ + Cs, C₁₇H₂₆N₆O₅ requires 527.0977).

Diastereoselective Reaction of the Stannous (Z)-Enolate of (4S, 5R)-3-((Methylthio)acetyl)-4-methyl-5-phenyl-2-oxazolidinone with 16. Tin(II) trifluoromethanesulfonate (1.6 mmol, 663 mg, 4.0 equiv) was dissolved in dry THF (2 mL) under N₂, cooled to -78 °C, and treated sequentially with (4S,5R)-3-((methylthio)acetyl)-4-methyl-5-phenyl-2oxazolidinone²³ (0.78 mmol, 211 mg, 2.0 equiv) in dry THF (2 mL) and iPr2NEt (1.75 mmol, 226 mg, 0.31 mL, 4.4 equiv). The mixture was stirred for 1 h at -20 °C for complete enolization, and the reaction mixture was recooled to -78 °C. A solution of 16 (0.40 mmol, 157 mg) in dry THF (2 mL) was added, and the reaction mixture was allowed to warm to 0 °C, where it was stirred for 12 h. The reaction mixture was poured into a two-layer solution of CH₂Cl₂ (30 mL) and saturated aqueous $NaHCO_3$ (15 mL) with vigorous stirring. The organic layer was washed with saturated aqueous NaCl (2×7 mL), dried (MgSO₄), and concentrated in vacuo. Chromatography (SiO₂, 3×6 cm, 5% CH₃. OH-CH₂Cl₂) gave 213 mg (262 mg theoretical, 81%) of a 82:5:13 mixture of three diastereomers (87:13 18a and 18b) determined to be the anti product (anti-18a), the corresponding syn product (syn. 18a), and third diastereomer, anti-18b, possessing the undesired 3R configuration. Upon prolonged reaction times, the ratio of the three products changed (24 h, 0 °C, 46:39:15), indicating the in situ epimerization of the C2 thiomethyl center, but the critical 18a:18b ratio remained unchanged (85-87:15-13). Preparative HPLC separation (SiO₂, 97:3 CHCl₃-CH₃OH) of the mixture of products afforded anti-18a (160 mg), syn-18a (10 mg), and the third diastereomer (anti-18b, 24 mg) as white solids.

Ethyl 2(R)-[1-[[2(S)-[((tert-Butyloxy)carbonyl)amino]-2-carbamoy]ethyl]amino]-2-[((4S,5R)-4-methyl-5-phenyl-2-oxazolidinyl)carbonyl]-2(R)-(methylthio)ethyl]-6-amino-5-methylpyrimidine-4-carboxylate (anti-18a). Mp 124-127 °C (EtOAc-hexane); Rf 0.34 (10% CH₃OH-CH₂Cl₂); $[\alpha]^{25}$ _D -27.8 (*c* 0.44, CHCl₃); ¹H NMR (CDCl₃, 400 MHz) δ 7.50–7.20 (5H, m), 6.97 (2H, br s), 5.81 (2H, br s), 5.75 (1H, d, J = 7.2 Hz), 5.74 (1H, d, J = 8.2 Hz), 4.94 (1H, d, J = 11.2 Hz), 4.84 (dq, J = 6.5, 7.2 Hz), 4.44 (2H, q, J = 7.1 Hz), 4.23 (1H, d, J = 11.2Hz), 3.98 (1H, br s), 3.08 (1H, dd, J = 3.1, 13.1 Hz), 2.62 (1H, br s), 2.25 (3H, s), 2.14 (3H, s), 1.41 (3H, t, J = 7.1 Hz), 1.36 (9H, s), 0.99 (3H, d, J = 6.5 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ 174.1, 169.5, 166.0, 165.2, 164.3, 155.7, 153.0, 152.7, 132.7, 128.6, 128.5, 125.5, 111.2, 79.3, 78.7, 64.1, 61.5, 55.0, 53.2, 49.2, 46.2, 27.9, 14.6, 13.9, 11.8, 11.3; IR (CHCl₃) v_{max} 3416, 2985, 1777, 1725, 1689, 1612, 1570, 1363, 1223, 1196, 1120, 1070 cm⁻¹; FABHRMS (NBA) m/e 660.2820 (M⁺ + H, C₃₀H₄₁N₇O₈S requires 660.2816).

Anal. Calcd for $C_{30}H_{41}N_7O_8S$: C, 54.59; H, 6.26; N, 14.87. Found: C, 54.39; H, 6.14; N, 14.52.

Ethyl 2(*R*)-[1-[[2(*S*)-[((*tert*-Butyloxy)carbonyl)amino]-2-carbamoylethyl]amino]-2-[((4*S*,5*R*)-4-methyl-5-phenyl-2-oxazolidinyl)carbonyl]-2(*S*) (methylthio)ethyl]-6-amino-5-methylpyrimidine-4-carboxylate (*sym*-18a). Mp 123-135 °C (EtOAc-hexane); *R* 0.30 (10% CH₃OH-CH₂Cl₂); $[\alpha]^{25}_{D}$ -6.5 (*c* 0.39, CHCl₃); ¹H NMR (CD₃OD, 400 MHz) δ 7.50-7.20 (5H, m), 5.69 (1H, d, *J* = 8.1 Hz), 5.29 (1H, d, *J* = 10.6 Hz), 4.65 (1H, dq, *J* = 6.5, 7.1 Hz), 4.35 (2H, q, *J* = 7.1 Hz), 2.88 (1H, dd, *J* = 6.7, 13.0 Hz), 2.80 (1H, dd, *J* = 5.6, 12.7 Hz), 2.11 (3H, s), 2.10 (3H, s), 1.41 (9H, s), 1.34 (3H, t, *J* = 7.1 Hz), 0.77 (3H, d, *J* = 6.5 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ 173.8, 169.7, 165.7, 165.6, 163.4, 155.6,

⁽⁴¹⁾ The amine displacement of the 2-(chloromethyl)pyrimidine provided 13 in a comparable conversion. In addition, 13 was prepared from imine 16 through catalytic hydrogenation (0.1 wt equiv of PtO_2 , H_2 , 25 °C, 16 h) and subsequent protection of the secondary amine (BOC₂O), but the sequence detailed in Scheme 2 proved superior.

152.1, 151.8, 132.9, 132.6, 128.2, 125.2, 110.3, 78.4, 78.2, 61.4, 60.5, 54.0, 53.2, 47.7, 46.1, 27.9, 14.3, 13.8, 11.4, 11.3; IR (CHCl₃) ν_{max} 3515, 3413, 1778, 1687, 1612, 1570, 1450, 1368, 1223, 1199, 1075 cm⁻¹; FABHRMS (NBA-CsI) *m/e* 792.1800 (M⁺ + Cs, C₃₀H₄₁N₇O₈S requires 792.1792).

Ethyl 2(S)-[1-[[2(S)-[((*tert*-Butyloxy)carbonyl)amino]-2·carbamoylethyl]amino]-2-[((4S,5R)-4-methyl-5-phenyl-2-oxazolidinyl)carbonyl]-2(S)-(methylthio)ethyl]-6-amino-5-methylpyrimidine-4-carboxylate (anti-18b). Mp 111-114 °C (EtOAc-hexane); R_f 0.32 (10% CH₃OH-CH₂Cl₂); $[\alpha]^{25}_{D}$ +14.8 (c 0.27, CHCl₃); ¹H NMR (CD₃OD, 400 MHz) δ 7.50-7.25 (5H, m), 5.87 (1H, d, J = 8.1 Hz), 5.05 (1H, d, J = 10.6 Hz), 4.94 (1H, dq, J = 6.5, 7.2 Hz), 4.42 (2H, q, J = 7.2 Hz), 4.15 (1H, d, J = 10.6 Hz), 4.04 (1H, m), 2.87 (1H, dd, J = 3.5, 13.2 Hz), 2.75 (1H, dd, J = 6.0, 13.1 Hz), 2.17 (3H, s), 2.02 (3H, s), 1.42 (3H, t, J = 7.2 Hz), 1.41 (9H, s), 0.81 (3H, d, J = 6.5 Hz); ¹³C NMR (CD₃OD, 100 MHz) δ 176.6, 171.0, 167.7, 167.5, 165.8, 157.5, 154.3, 153.8, 135.2, 129.8, 129.5, 127.0, 112.0, 80.5, 80.2, 65.4, 63.0, 55.7, 55.3, 51.0, 48.5, (88.7, 14.6, 14.5, 12.2, 12.0; IR (CHCl₃) ν_{max} 3403, 2993, 1778, 1723, 1687, 1612, 1573, 1452, 1364, 1221, 1198, 1071 cm⁻¹; FABHRMS (NBA-CsI) m/e 792.1760 (M⁺ + Cs, C₃₀H₄₁N₇O₈S requires 792.1792).

Ethyl 2(S)-[1-[[2(S)-[((tert-Butyloxy)carbonyl)amino]-2-carbamoylethyl]amino]-2-[((4S,5R)-4-methyl-5-phenyl-2-oxazolidinyl)carbonyl]ethyl]-6-amino-5-methylpyrimidine- 4-carboxylate (19). A solution of anti-18a (0.29 mmol, 192 mg) in C_6H_6 (2 mL) was treated with Bu_3SnH (0.87 mmol, 250 mg, 3.0 equiv) and AIBN (0.042 mmol, 7 mg, 0.15 equiv) and was warmed at 80 °C (45 min) under N2. The mixture was allowed to cool to 23 °C, and the solvent was evaporated in vacuo. Flash chromatography (SiO₂, 3 × 10 cm, 5% CH₃OH-CH₂Cl₂) afforded 19 as a white solid (162 mg, 178 mg theoretical, 91%): mp 122-124 °C (EtOAc-hexane); $R_f 0.28$ (10% CH₃OH-CH₂Cl₂); $[\alpha]^{25}D$ -19.3 (c 0.29, CHCl₃); ¹H NMR (CDCl₃, 400 MHz) δ 7.75 (1H, br s), 7.43 (1H, m), 7.40–7.20 (5H, m), 6.01 (1H, br s), 5.74 (2H, br s), 5.67 (1H, d, J = 7.2 Hz, 5.58 (1H, br s), 5.47 (1H, m), 4.76 (1H, dq, J = 6.6, 7.2 Hz), 4.41 (2H, q, J = 7.1 Hz), 4.27 (1H, m), 4.02 (1H, m), 3.47 (1H, dd, J= 8.4, 16.0 Hz), 3.28 (1 H, d, J = 14.8 Hz), 3.15 (1 H, d, J = 11.8 Hz),2.62 (1H, dd, J = 7.0, 13.1 Hz), 2.19 (3H, s), 1.43 (9H, s), 1.40 (3H, t, J = 7.1 Hz), 0.90 (3H, d, J = 6.6 Hz); ¹³C NMR (CD₃OD, 100 MHz) δ 176.2, 172.0, 168.8, 167.4, 165.6, 157.5, 154.5, 154.2, 135.0, 129.4, 129.2, 126.9, 111.1, 80.5, 80.1, 62.8, 61.0, 55.7, 55.2, 49.7, 41.4, 28.4, 14.7, 14.2, 11.8; IR (CHCl₃) v_{max} 3415, 3010, 1780, 1689, 1611, 1569, 1493, 1370, 1223, 1197, 1069 cm⁻¹; FABHRMS (NBA-CsI) m/e 746.1914 (M^+ + Cs, C₂₉H₃₉N₇O₈ requires 746.1914).

Anal. Calcd for $C_{29}H_{39}N_7O_8$: C, 56.74; H, 6.41; N, 15.98. Found: C, 56.72; H, 6.31; N, 15.47.

Similar treatment of syn-18a provided 19, identical in all respects. N^a-((tert-Butyloxy)carbonyl)-N³-[1-amino-3(S)-(6-amino-4-(ethoxycarbonyl)-5-methylpyrimidin-2-yl)propion-3-yl]-(S)-\$-aminoalanine Amide (20). Solid 19 (0.16 mmol, 99 mg) was treated with an ethanolic solution of NH₃ (16%, 20 mL), and the solution was stirred for 1.5 h at 0 °C. The solvent was evaporated in vacuo. Flash chromatography (SiO₂, 3×5 cm, 10% CH₃OH-CH₂Cl₂) afforded 20 as a white solid (59 mg, 73 mg theoretical, 80%): mp 157-159 °C (iPrOH-hexane) (lit4e mp 159-162 °C (*i*PrOH)); R_f 0.32 (20% CH₃OH-CHCl₃); $[\alpha]^{25}D$ -10.8 (c 0.36, EtOH) ($lit^{4a} [\alpha]^{25} - 7.5 (c 1.0, EtOH)$); ¹H NMR (CD₃OD, 400 MHz) δ 4.31 (2H, q, J = 7.1 Hz), 4.03 (1H, m), 3.88 (1H, dd, J = 4.9, 8.8 Hz), 2.69 (2H, m), 2.53 (1H, dd, J = 5.0, 15.0 Hz), 2.45 (1H, dd, J = 8.9, 15.0 Hz), 2.04 (3H, s), 1.35 (9H, s), 1.29 (3H, t, J = 7.1 Hz); ¹³C NMR (CD₃OD, 100 MHz) § 176.5, 176.4, 168.3, 167.7, 165.9, 157.8, 154.4, 111.5, 80.8, 63.0, 62.0, 55.7, 50.0, 42.0, 28.7, 14.5, 12.1; IR (neat) ν_{max} 3416, 2978, 1727, 1678, 1611, 1517, 1423, 1214, 1046, 929 cm⁻¹; FABHRMS (NBA-CsI) m/e 586.1396 (M+ + Cs, C19H31N7O6 requires 586.1390)

N²-((*tert*-Butyloxy)carbonyl)-N²-[1-amino-3(S) ·(6-amino-4-carboxy-5-methylpyrimidin-2-yl)propion-3-yl]-(S)-β-aminoalanine Amide (21). A solution of 20 (0.046 mmol, 20.7 mg) in THF-CH₃OH-H₂O (3:1:1, 0.5 mL) at 0 °C was treated with aqueous 1 N LiOH (0.07 mmol, 0.07 mL, 1.5 equiv), and the mixture was stirred for 1.5 h. After evaporation of most of the THF-CH₃OH, the aqueous phase was acidified to pH 4-5 with the addition of aqueous 1.2 N HCl and the solvent was evaporated in vacuo. The residue was charged onto a column of Dowex (1 × 8 cm, acetate form, 50-100 mesh). The column was washed with H₂O, and subsequent elution with 6% HOAc-H₂O afforded 21 as a white powder (17.6 mg, 19 mg theoretical, 91%): mp 220-222 °C (EtOH-hexane) (lit⁴⁴ mp 220-222 °C (EtOH)); R_f 0.58 (4:1:1 *i*PrOH-H₂O-HOAc); [α]²⁵_D -35.6 (c 0.81, H₂O) (lit [α]²⁸_D -33.6⁴⁴ and -32.8⁴⁶ (c 0.75, H₂O)); ¹H NMR (D₂O, 400 MHz) δ 4.17 (2H, m), 3.16 (1H, m), 3.03 (1H, m), 2.76 (2H, m), 1.93 (3H, s), 1.30 (9H, s); ¹³C NMR (CD₃OD, 100 MHz) δ 174.6, 169.0, 166.5, 161.1, 157.6, 154.2, 112.9, 81.6, 60.6, 52.1, 48.5, 36.3, 28.6, 12.2; IR (neat) ν_{max} 3423, 3189, 1720, 1685, 1479, 1258, 1161, 1022, 878 cm⁻¹; FABHRMS (NBA–CsI) *m/e* 426.2118 (M⁺ + H, C₁₇H₂₇N₇O₆ requires 426.2101).

(-)-Pyrimidoblamic Acid (1). The solid 21 (0.005 mmol, 2.1 mg) was treated with 3 N HCl-EtOAc (0.5 mL), and the mixture was stirred at 25 °C for 1 h. The solvent was evaporated in vacuo to give pure 1 hydrochloride (2.2 mg, 2.2 mg theoretical, 100%) as a clear, hygroscropic solid: R_f 0.32 (4:1:1 iPrOH-H₂O-HOAc); $[\alpha]^{25}_{D}$ -26.7 (c 0.12, H₂O); ¹H NMR (D₂O, 400 MHz) δ 4.24 (1H, dd, J = 7.9, 7.9 Hz), 4.12 (1H, dd, J = 4.5, 6.6 Hz), 3.21 (1H, dd, J = 4.5, 13.7 Hz), 3.08 (1H, dd, J = 6.6, 13.7 Hz), 2.89 (1H, dd, J = 5.4, 15.8 Hz), 2.82 (1H, dd, J = 7.4, 15.8 Hz), 2.22 (3H, s); IR (neat) ν_{max} 3456, 3247, 1695, 1681, 1557, 1161, 1078, 820 cm⁻¹; FABHRMS (NBA) m/e 326.1564 (M⁺ + H, C₁₂H₁₉N₇O₄ requires 326.1577).

Ethyl 2(R)-[1-[[2(S)-[((tert-Butyloxy)carbonyl)amino]-2-carbamoylethyl]amino]-2-[((45,5R)-4-methyl-5-phenyl-2-oxazolidinyl)carbonyl]ethyl]-6-amino-5-methylpyrimidine-4-carboxylate (22). A solution of 18b (0.09 mmol, 58 mg) in C_6H_6 (2 mL) was treated with Bu_3SnH (0.26 mmol, 77 mg, 3 equiv) and AIBN (6.02 mmol, 3 mg, 0.15 equiv), and the solution was warmed at 80 °C (45 min) under N_2 . The mixture was allowed to cool to 23 °C, and the solvent was evaporated in vacuo. Flash chromatography (SiO₂, 3 × 3 cm, 5% CH₃OH-CH₂Cl₂) afforded 22 (49 mg, 53 mg theoretical, 92%) as a white solid: mp 121-123 °C (EtOAchexane); $R_f 0.28$ (10% CH₃OH-CH₂Cl₂); $[\alpha]^{25}D$ -1.1 (c 0.18, CHCl₃); ¹H NMR (CDCl₃, 400 MHz) § 7.40-7.20 (6H, m), 6.40 (1H, m), 5.79 (2H, br s), 5.71 (1H, d, J = 7.2 Hz), 5.50 (2H, m), 4.77 (1H, dq, J = 6.6, 7.2 Hz), 4.42 (2H, q, J = 7.2 Hz), 4.35 (1H, dd, J = 5.5, 7.5 Hz), 4.11 (1H, m), 3.39 (2H, m), 3.16 (1H, m), 2.71 (1H, dd, J =6.5, 13.0 Hz), 2.18 (3H, s), 1.45 (9H, s), 1.40 (3H, t, J = 7.2 Hz), 0.91 $(3H, d, J = 6.6 \text{ Hz}); {}^{13}\text{C} \text{ NMR} (CD_3\text{OD}, 100 \text{ MHz}) \delta 176.4, 172.2,$ 168.1, 167.5, 165.6, 157.5, 154.6, 154.3, 135.1, 129.5, 129.3, 126.9, 111.1, 80.5, 80.1, 62.8, 60.9, 55.7, 55.5, 49.8, 41.6, 32.5, 14.6, 14.3, 11.9; IR (CHCl₃) v_{max} 3418, 3018, 2981, 1729, 1687, 1611, 1573, 1455, 1368, 1215, 1122, 1069, 780 cm-1; FABHRMS (NBA-CsI) m/e 746.1944 (M+ + Cs, C₂₉H₃₉N₇O₈ requires 746.1914).

Nⁿ-((*tert*-Butyloxy)carbonyl)-Nⁿ-[1-amino-3(R)-(6-amino-4-(ethoxy-carbonyl)-5-methylpyrimidin-2-yl)propion-3-yl]-(S)-β-aminoalanine Amide (23). Compound 22 (0.064 mmol, 39 mg) was subjected to aminolysis as described for 19 to provide 23 (24 mg, 29 mg theoretical, 82%) as a white solid: mp 69-72 °C (*l*PrOH-hexane) (lit^{4a} mp 64-66 °C (*l*PrOH)); R_f 0.32 (20% CH₃OH-CHCl₃); $[\alpha]^{25}_D$ +38 (c 0.57, EtOH) (lit^{4a} $[\alpha]^{25}_D$ +14.8 (c 1.0, EtOH)); 1H NMR (CD₃OD, 400 MHz) δ 4.31 (2H, q, J = 7.1 Hz), 4.01 (1H, m), 3.88 (1H, dd, J = 5.3, 8.3 Hz), 2.75 (1H, dd, J = 5.3, 12.4 Hz), 2.68 (1H, dd, J = 6.4, 10.4 Hz), 2.47 (2H, m), 2.04 (3H, s), 1.33 (9H, s), 1.29 (3H, t, J = 7.1 Hz); ¹³C NMR (CD₃OD, 100 MHz) δ 176.7, 168.4, 167.8, 165.9, 157.8, 154.4, 111.4, 80.7, 63.0, 61.9, 55.8, 50.0, 41.8, 28.7, 14.5, 12.1; IR (neat) ν_{max} 3407, 2996, 1725, 1653, 1609, 1519, 1474, 1420, 1215, 1047, 927, 880 cm⁻¹; FABHRMS (NBA-CsI) m/e 586.1396 (M⁺ + Cs, C₁₉H₃₁N₇O₆ requires 586.1390).

N^{*n*}-((*tert*-Butyloxy)carbonyl)-N^{*θ*}-[1-amino-3(*R*)-(6-amino-4-carboxy-5-methylpyrimidin-2-yl)propion-3-yl]-(S)-β-aminoalanine Amide (24). Compound 23 (0.022 mmol, 9.9 mg) was subjected to hydrolysis and purification as described for 20 to provide 24 (8.8 mg, 9.3 mg theoretical, 94%): mp 221-223 °C (EtOH-hexane) (lit^{4a} mp 221 °C); *R_f* 0.58 (4: 1:1, *i*PrOH-H₂O-HOAC); $[\alpha]^{25}_{D}$ +20.8 (*c* 0.44, H₂O) (lit^{4a} $[\alpha]^{25}_{D}$ +20.8 (*c* 0.65, H₂O)); ¹H NMR (D₂O, 400 MHz) δ 4.08 (2H, m), 3.06 (2H, m), 2.76 (2H, m), 2.00 (3H, s), 1.30 (9H, s); ¹³C NMR (CD₃OD, 100 MHz) 174.5, 174.3, 168.4, 166.5, 160.9, 157.6, 152.9, 113.9, 81.7, 61.0, 52.0, 48.5, 35.7, 28.6, 12.2; IR (neat) ν_{max} 3412, 3145, 1717, 1686, 1584, 1479, 1370, 1258, 1160, 873 cm⁻¹; FABHRMS (NBA) *m/e* 426.2118 (M⁺ + H, C₁₇H₂₇N₇O₆ requires 426.2101).

epi-(+)-Pyrimidoblamic Acid (2). The solid 24 (0.005 mmol, 2.2 mg) was subjected to acid-catalyzed deprotection in the same fashion as 21 to give epi-(+)-pyrimidoblamic acid hydrochloride (2, 2.4 mg, 2.4 mg theoretical, 100%) as a clear, hygroscopic solid: $R_f 0.32$ (4:1:1 *i*PrOH- H_2O -HOAc;); [α]²⁵_D+20.1 (c 0.11, H_2O); ¹H NMR (D₂O, 400 MHz) δ 4.25 (1H, dd, J = 6.6, 6.6 Hz), 4.10 (1H, dd, J = 5.8, 5.8 Hz), 3.18 (1H, dd, J = 5.4, 13.5 Hz), 3.12 (1H, dd, J = 6.4, 13.6 Hz), 2.90 (1H, dd, J = 5.8, 16.6 Hz), 2.82 (1H, dd, J = 7.1, 16.0 Hz), 2.21 (3H, s); IR (neat) ν_{max} 3448, 3256, 1692, 1682, 1468, 1161, 1021, 810 cm⁻¹; FABHRMS (NBA) *m/e* 326.1567 (M⁺ + H, Cl₂H₁₉N₇O₄ requires 326.1577).

Carbamate 25.42 A solution of anti-18s (52 mg, 0.079 mmol) in anhydrous CH₃OH (1 mL) was cooled to 0 °C and treated with NaBH₄ (20 mg, 0.55 mmol, 7 equiv) under Ar. After being stirred at 0 °C for 1 h, the reaction mixture was quenched with the addition of H_2O (0.2 mL). The mixture was extracted with 10% CH₃OH-CH₂Cl₂ (2 \times 5 mL), and the combined organic extracts were dried (Na₂SO₄) and concentrated in vacuo. Flash chromatography (SiO₂, 3×5 cm, 5% CH₃OH-CH₂Cl₂) gave the corresponding alcohol as a white foam. A solution of the alcohol (0.011 mmol, 4.5 mg) in anhydrous CH_2Cl_2 (0.2 mL) was treated with pyridine (20 mg, 0.25 mmol) and COCl₂ (0.087 mmol, 45 µL, 1.93 M in toluene) at 25 °C. After being stirred at 25 °C for 30 min, the reaction mixture was poured into a two-layer solution of CH2Cl2 and saturated aqueous NaHCO3. The organic layer was washed with 10% aqueous HCl, saturated aqueous NaHCO₃, and saturated aqueous NaCl, dried (Na₂SO₄), and concentrated in vacuo. Flash chromatography (SiO₂, 3×5 cm, 5% CH₃OH-CH₂Cl₂) gave 25 (1.7 mg, 4.3 mg theoretical, 40%) as a white foam: $R_f 0.33$ (10% CH₃OH-CHCl₃); [α]²⁵_D+14.5 (c 0.06, CHCl₃); ¹H NMR (CD₃OD, 400 MHz) δ 4.78 (1H, m), 4.59 (1H, d, J = 1.5 Hz, C4-H), 4.47 (1H, dd, J = 2.0, 11.0 Hz), 4.30 (2H, q, J = 7.2 Hz), 4.08 (1H, dd, J = 1.5, 11.0 Hz), 3.75 (1H, dd, J = 7.3, 13.5 Hz), 3.38 (1H, br s), 3.21 (1H, m), 2.18 (3H, s),2.05 (3H, s), 1.33 (9H, s), 1.29 (3H, t, J = 7.2 Hz); IR (CHCl₃) ν_{max} 3010, 2976, 2399 (s, C=N), 1734, 1700, 1521, 1476, 1423, 1215, 1046, 928, 876, 774, 669 cm⁻¹; FABHRMS (NBA-CsI) m/e 627.1002 (M⁺ + C_{s} , $C_{21}H_{30}N_6O_6S$ requires 627.1002).

Carbamate 26.⁴² syn-18a (0.014 mmol, 9.1 mg) was subjected to reduction and purification as described for 25 to provide the corresponding alcohol (4.0 mg, 6.6 mg theoretical, 60%) as a white foam. The alcohol (0.009 mmol, 4.0 mg) was subjected to carbamate formation in the same fashion as 25 to give 26 (1.8 mg, 4.5 mg theoretical, 42%): R_f 0.32 (10% CH₃OH-CHCl₃); $[\alpha]^{25}_D$ +18.0 (c 0.05, CHCl₃); ¹H NMR (CD₃-OD, 400 MHz) δ 4.77 (1H, m), 4.66 (1H, d, J = 5.6 Hz, C4-H), 4.32 (2H, q, J = 7.2 Hz), 4.21 (1H, dd, J = 4.9, 10.3 Hz), 3.75 (1H, dd, J = 8.2, 10.3 Hz), 3.53 (1H, dd, J = 4.1, 13.0 Hz), 3.48 (1H, m), 3.01 (1H, dd, J = 8.3, 13.0 Hz), 2.07 (3H, s), 2.05 (3H, s), 1.34 (9H, s), 1.31 (3H, t, J = 7.2 Hz); IR (CHCl₃) ν_{max} 3010, 2975, 2399 (s, C=N), 1732, 1699, 1520, 1476, 1423, 1210, 1045, 928 cm⁻¹; FABHRMS (NBA-CsI) m/e 627.1021 (M⁺ + Cs, C₂₁H₃₀N₆O₆S requires 627.1002).

Carbamate 27.⁴² anti-18b (0.048 mmol, 32 mg) was subjected to reduction and purification as described for 25 to provide the corresponding alcohol as a white foam. The alcohol (0.009 mmol, 4.2 mg) was subjected to carbamate formation in the same fashion as 25 to provide 27 (3.2 mg, 4.1 mg theoretical, 78%): ¹H NMR (CD₃OD, 400 MHz) δ 4.80 (1H, m), 4.72 (1H, d, J = 1.2 Hz, C4·H), 4.46 (1H, dd, J = 1.8, 12.0 Hz), 4.33 (2H, q, J = 7.2 Hz), 4.07 (1H, dd, J = 1.5, 12.0 Hz), 3.87 (1H, dd, J = 4.5, 14.0 Hz), 3.37 (1H, br s), 3.05 (1H, dd, J = 4.8, 140 Hz), 2.17 (3H, s), 2.06 (3H, s), 1.38 (9H, s), 1.28 (1H, t, J = 7.2 Hz).

Methyl 3(5)-(6-Amino-4-(methoxycarbonyl)-5-methylpyrinidin-2-yl)-3-aminopropionate (31). From 20: Compound 20 (0.011 mmol, 5.2 mg) was treated with 6 N HCl (0.7 mL), and the mixture was stirred at 105 °C for 20 h. The solvent was evaporated in vacuo and the oily solid was treated with 3 N HCl-CH₃OH (1 mL), and the mixture was stirred at 80 °C for 1 h. After evaporation of the solvent, the residue was subjected to chromatography (SiO₂, 1 × 2 cm, 20% CH₃OH-CHCl₃) to afford 31 (2.2 mg, 2.9 mg theoretical, 75%): R_f 0.32 (25% CH₃OH-CHCl₃); $[\alpha]^{32}$ +3.4 (c 0.15, 1 N HCl), $[\alpha]^{25}_{365}$ +36.7 (c 0.15, 1 N HCl); ¹H NMR (CD₃OD, 400 MHz) δ 4.66 (1H, dd, J = 6.4, 7.2 Hz), 3.96 (3H, s), 3.71 (3H, s), 3.16 (1H, dd, J = 6.4, 16.4 Hz), 3.10 (1H, dd, J = 7.2, 16.4 Hz), 2.18 (3H, s); IR (neat) ν_{max} 3445, 1715, 1630, 1214, 1143 cm⁻¹; FABHRMS (NBA) m/e 269.1255 (M⁺ + H, C₁₁H₁₆N₄O₄ requires 269.1250).

From bleomycin A₂: A solution of bleomycin A₂ (0.043 mmol, 5.0 mg) was subjected to the same conditions detailed above to provide 31 (0.5 mg, 1.2 mg theoretical, 41%) identical to the material detailed above: $[\alpha]^{25}_{365} + 26$ (c 0.15, 1 N HCl).

3(S)-(6-Amino-4-carboxy 5-methylpyrimldin $2 \cdot yi$) $\cdot 3 \cdot amino$ propionic Acid (30). The compound 31 (0.007 mmol, 1.8 mg, from 20) was treated with aqueous 3 N HCl (0.3 mL), and the mixture was stirred at 100 °C for 1 h. The solvent was evaporated in vacuo to afford **30** (1.5 mg, 1.6 mg theoretical, 94%): $[\alpha]^{25}_{365}$ +39.6 (c 0.07, 1 N HCl); ¹H NMR (D₂O, 400 MHz) δ 4.94 (1H, m), 3.24 (1H, dd, J = 5.8, 16.2 Hz), 3.10 (1H, dd, J = 9.0, 16.2 Hz), 2.25 (3H, s).

The sample of **31** derived from bleomycin A₂ similarly provided **30**: $[\alpha]^{23}_{365}$ +30 (c 0.2, 1 N HCl).

Methyl 3(*R*)-(6-Amino-4-(methoxycarbonyl)-5-methylpyrimidin-2-yl)-3-aminopropionate (32). Compound 23 (0.012 mmol, 5.7 mg) was subjected to chemical degradation followed by esterification as described for 2D and bleomycin A₂ to provide 32 (2.1 mg, 3.3 mg theoretical, 64%) identical to 31 except for rotation: $[\alpha]^{25}D-3.7 (c 0.14, 1 N HCl), [\alpha]^{25}_{365}$ -39.3 (c 0.14, 1 N HCl).

3(*R*)-(6-Amino-4 carboxy-5-methylpyrlmldln-2-yl)·3-aminopropionic Acid (33). Compound 32 (0.0075 mmol, 2.0 mg) was subjected to hydrolysis as described for 31 to provide 33 (1.7 mg, 1.8 mg theoretical, 94%) identical to 30 except for rotation: $[\alpha]^{25}_{365}$ -40.3 (c 0.08, 1 N HCl).

Subjection of 39 to the same hydrolysis conditions also provided 33 identical to that detailed above but of higher enantiomeric purity: $[\alpha]^{25}_{365}$ -59 (c 0.11, 1 N HCl).

Ethyl 2-[1(R)·Hydroxy-2(S)·(methylthio)-2·[((4S,5R)-4-methyl-5phenyl-2-oxazolidinyl)carbonyl]ethyl]-6-amino-5-methylpyrimidine-4-carboxylate (35). A solution of $(4S, 5R) \cdot 3$ -((methylthio)acetyl)-4-methyl-5-phenyl-2-oxazolidinone (34, 0.15 mmol, 38 mg) in dry CH₂Cl₂ (0.7 mL) under N_2 at -78 °C was treated with Bu_2BOTf (0.15 mmol, 41 mg, 33 µL, 1.0 equiv) followed by iPr2NEt (0.165 mmol, 21 mg, 29 µL, 1.1 equiv). The mixture was stirred at 0 °C for 1 h and recooled to -78 °C. A solution of 15 (0.038 mmol, 7.8 mg, 0.25 equiv) in dry CH₂Cl₂ (0.2 mL) was added, and the reaction mixture was allowed to warm to 0 °C, where it was stirred for 3 h. The reaction mixture was poured into a two-phase solution of CH₂Cl₂ (2 mL) and saturated aqueous NaHCO₃ (2 mL) with vigorous stirring. The organic layer was washed with saturated aqueous NaCl (2×1 mL), dried over (Na₂SO₄), and concentrated in vacuo. Chromatography (SiO₂, 1×4 cm, 2% CH₃. $OH-CH_2Cl_2$) gave 35 as a white foam (10 mg, 18 mg theoretical, 59%): $R_f 0.32$ (10% CH₃OH-CH₂Cl₂); $[\alpha]^{25}D$ -8.0 (c 0.52, CHCl₃); ¹H NMR (CDCl₃, 400 MHz) δ 7.40–7.20 (5H, m), 5.75 (1H, d, J = 7.2 Hz), 5.30 (2H, br s), 5.26 (1H, d, J = 8.8 Hz), 5.13 (1H, dd, J = 4.0, 8.8 Hz),4.83 (2H, q, J = 6.8 Hz), 4.39 (2H, m), 2.23 (3H, s), 2.20 (3H, s), 1.40 $(3H, t, J = 6.0 \text{ Hz}), 0.92 (3H, d, J = 6.8 \text{ Hz}); \text{ IR (CHCl}_3) \nu_{\text{max}} 3419,$ 3018, 1778, 1729, 1690, 1612, 1570, 1521, 1443, 1368, 1213, 1671, 928, 756 cm⁻¹; FABHRMS (NBA-CsI) m/e 607.0632 (M⁺ + Cs, C₂₂H₂₆N₄O₆S requires 607.0627).

Ethyl 2-[1(S)-Hydroxy-2.[((4S,5R).4-methyl-5-phenyl-2-oxazoli. dinyl)carbonyl]ethyl]-6-amino-5-methylpyrimidine-4-carboxylate (36). A solution of 35 (0.044 mmol, 20 mg) in C_6H_6 (0.7 mL) was treated with Bu₃SnH (0.13 mmol, 38 mg, 3.0 equiv) and AIBN (0.013 mmol, 3 mg, 0.3 equiv) and warmed at 80 °C for 1 h under N2. The mixture was allowed to cool to 23 °C, and the solvent was evaporated in vacuo. Flash chromatography (SiO₂, 1×4 cm, 5% CH₃OH-CH₂Cl₂) afforded 36 as a white foam (15.6 mg, 17 mg theoretical, 93%): $R_f 0.32$ (10% CH₃-OH-CH₂Cl₂); [a]²⁵_D -37.2 (c 0.42, CHCl₃); ¹H NMR (CDCl₃, 400 MHz) δ 7.40–7.20 (5H, m), 5.91 (2H, br s), 5.70 (1H, d, J = 7.2 Hz), 5.15 (1H, dd, J = 3.3, 8.0 Hz), 4.80 (1H, dq, J = 6.8, 7.2 Hz), 4.41 (2H, J = 6.8, 7q, J = 6.8 Hz, 3.61 (1H, dd, J = 3.6, 16.4 Hz), 3.47 (1H, dd, J = 8.0, J = 8.016.4 Hz), 2.19 (3H, s), 1.39 (3H, t, J = 6.8 Hz), 0.90 (1H, d, J = 6.8Hz); IR (CHCl₃) v_{max} 3420, 3018, 1780, 1726, 1612, 1521, 1424, 1214, 1043, 928 cm⁻¹; FABHRMS (NBA-CsI) m/e 561.0747 (M⁺ + Cs, $C_{21}H_{24}N_4O_6$ requires 561.0750).

Ethyl 2-[1(\mathbb{R}) Azido-2-[((4S, 5 \mathbb{R})-4-methyl-5-phenyl-2-oxazolidinyl)carbonyl]ethyl]-6-amino-5-methylpyrimidine-4-carboxylate (37). Diethyl azodicarboxylate (0.043 mmol, 7.5 mg, 7.0 μ L, 1.2 equiv) and a 0.8 M benzene solution of hydrazoic acid⁴³ (0.043 mmol, 54 μ L, 1.2 equiv) were added sequentially to a solution of **36** (0.036 mmol, 15 mg) and Ph₃P (0.043 mmol, 12 mg, 1.2 equiv) in dry THF (0.5 mL) at -78 °C. The mixture was stirred for 16 h at 23 °C. The solvent was evaporated in vacuo. Flash chromatography (SiO₂, 1 × 6 cm, 2% CH₃OH-CH₂Cl₂) afforded **37** as a foam (8.2 mg, 15.7 mg theoretical, 52%): \mathbb{R}_F o.042 (5% CH₃OH-CH₂Cl₂); [α]²⁵D+35.6 (c 0.53, CHCl₃); ¹H NMR (CDCl₃, 400 MHz) δ 7.30-7.10 (5H, m), 5.65 (1H, d, J = 7.2 Hz), 5.14 (2H, br s), 5.00 (1H, dd, J = 4.8, 8.4 Hz), 4.75 (1H, dq, J = 6.8, 7.2 Hz), 4.37 (2H, q, J = 7.2 Hz), 3.53 (1H, d, J = 7.2 Hz), 0.83 (3H, d, J = 6.8 Hz); IR (CHCl₃) ν_{max} 3018, 2107, 1723, 1709, 1611, 1519, 1473, 1422, 1388,

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⁽⁴²⁾ Reduction with excess NaBH₄ (EtOH) at 25 °C led to C4 ethyl ester reduction. Initial attempts to form the cyclic carbamates upon reaction with diethyl carbonate (K_2CO_3) and methyl chloroformate (C_6H_6 , pyridine, 25 °C, 30 min, 79%) provided corresponding uncyclized methyl carbamate which failed to close to 25–27 upon further treatment with base (DMAP, K_2CO_3 , DBU, NaH; C_6H_6 , DMF, THF; 25–80 °C). Similarly, treatment with carbonyldiimidazole ($C_6H_6, 80$ °C) provided the *N*-carbonyldiimidazole, which failed to close cleanly to the desired cyclic carbamate.

Synthesis of Pyrimidoblamic Acid and Analogs

1218, 1045, 928, 757 cm⁻¹; FABHRMS (NBA) m/e 454.1830 (M⁺ + H, C₂₁H₂₃N₇O₅ requires 454.1839).

Methyl3-(6-Amino-4-(ethoxycarbonyl).5-methylpyrimidin-2-yl)-3(R)azidopropionate (38). A solution of 37 (0.018 mmol, 8.0 mg) in CH₃OH (0.3 mL) under N₂ at 0 °C was treated with NaOCH₃ (0.018 mmol, 1.0 mg, 1.0 equiv), and the mixture was stirred at 0 °C for 20 min. The reaction mixture was poured into a two-layer solution of CH₂Cl₂ (2 mL) and saturated aqueous NaCl (1 mL) with vigorous stirring. The organic layer was washed with saturated aqueous NaCl (1 mL), dried (Na2SO4), and concentrated in vacuo. Chromatography (SiO₂, 1×3 cm, 2% CH₃-OH-CH₂Cl₂) gave 38 as a film (3.2 mg, 5.3 mg theoretical, 59%): R_f $0.35 (20\% \text{ CH}_{3}\text{OH}-\text{CH}_{2}\text{Cl}_{2}); [\alpha]^{25} + 61.6 (c 0.19, \text{CHCl}_{3}); ^{1}\text{H NMR}$ $(CDCl_3, 400 \text{ MHz}) \delta 5.13 (2H, \text{ br s}), 4.90 (1H, dd, J = 5.2, 8.8 \text{ Hz}),$ 4.42 (2H, q, J = 6.8 Hz), 3.73 (3H, s), 3.04 (1H, dd, J = 5.2, 17.2 Hz), 2.87 (1H, dd, J = 8.8, 17.2 Hz), 2.21 (3H, s), 1.41 (3H, t, J = 6.8 Hz);IR (CHCl₃) v_{max} 3017, 2392, 1722, 1508, 1364, 1242, 1208, 1046, 924, 843, 745 cm⁻¹; FABHRMS (NBA) m/e 309.1310 (M⁺ + H, C₁₂H₁₆N₆O₄ requires 309.1311).

Methyl 3-(6-Amino-4-(ethoxycarbonyl)-5-methylpyrimidin-2-yl)·3(**R**)**aminopropionate (39).** A solution of **38** (0.008 mmol, 25 mg) in THF (0.3 mL) was treated with Ph₃P (0.01 mmol, 4.5 mg, 2.0 equiv) in the presence of H₂O (5 μ L), and HOAc (5 μ L) and the mixture was stirred under N₂ at 25 °C for 15 h. The solvent was evaporated in vacuo. Flash chromatography (SiO₂, 1 × 2 cm, 20% CH₃OH-CH₂Cl₂) afforded **39** as a white film (1.9 mg, 2.3 mg theoretical, 82%): R_f 0.12 (5% CH₃-OH-CHCl₃); [α]²⁵_D-5.6 (c 0.16, 1 N HCl), [α]²⁵₃₆₅-53.8 (c 0.16, 1 N HCl); ¹H NMR (CD₃OD, 400 MHz) δ 4.77 (1H, br s), 4.27 (2H, q, J = 7.2 Hz), 3.49 (3H, s), 3.00 (1H, dd, J = 5.6, 17.4 Hz), 2.87 (1H, dd, J = 7.2, 17.4 Hz), 2.19 (3H, s), 1.19 (3H, t, J = 7.2 Hz); IR (CHCl₃) ν_{max} 3329, 2936, 1721, 1653, 1450, 1414, 1213, 1023, 924, 739 cm⁻¹; FABHRMS (NBA) m/e 283.1400 (M⁺ + H, C₁₂H₁₈N₄O₄ requires 283.1406).

 $3(R) \cdot (6-Amino \cdot 4-carboxy-5-methylpyrimidin \cdot 2 \cdot yl)-3-amino$ propionic Acid (33). The compound 39 (0.0087 mmol, 2.4 mg) wastreated with 3 N aqueous HCl (0.3 mL), and the mixture was stirred at100 °C for 1 h. The solvent was evaporated in vacuo to afford 33 (2.0 $mg, 2.2 mg theoretical, 91%): <math>[\alpha]^{25}_{365}-59.3$ (c 0.11, 1 N HCl); ¹H NMR (D₂O, 400 MHz) δ 4.94 (1H, m), 3.24 (1H, dd, J = 5.2, 17.1 Hz), 3.13 (1H, dd, J = 7.2, 17.1 Hz), 2.25 (3H, s); IR (neat) ν_{max} 3438, 2921, 1708, 1646, 1527, 1441, 1216, 1051, 865 cm⁻¹; FABHRMS (NBA) m/e 241.0930 (M⁺ + H, C₉H₁₂N₄O₄ requires 241.0937).

Ethyl 2-[1(S)-Hydroxy-2(R) (methylthio)-2-[((4R,5S) 4-methyl 5phenyl-2-oxazolidinyl)carbonyl]ethyl]-6-amino-5-methylpyrimidine-4-carboxylate (41). A solution of (4R,5S)-3-((methylthio)acetyl)-4-methyl-5-phenyl-2-oxazolidinone (40, 2.30 mmol, 576 mg) in dry CH₂Cl₂ (2 mL) under N₂ at -78 °C was treated with Bu₂BOTf (2.30 mmol, 630 mg, 1.0 equiv) followed by iPr2NEt (2.52 mmol, 326 mg, 0.44 mL, 1.1 equiv). The mixture was stirred at 0 °C for 1 h, and the reaction mixture was recooled to -78 °C. A solution of 15 (0.57 mmol, 120 mg, 0.25 equiv) in dry CH₂Cl₂ (2 mL) was added, and the reaction mixture was allowed to warm to 0 °C where it was stirred for 3 h. The reaction mixture was poured into a two-phase solution of CH2Cl2 (20 mL) and saturated aqueous NaHCO₃ (10 mL) with vigorous stirring. The organic layer was washed with saturated aqueous NaCl (2×5 mL), dried (Na₂· SO_4), and concentrated in vacuo. Chromatography (SiO₂, 1 × 8 cm, 2% CH₃OH-CH₂Cl₂) gave 41 as a thin film (180 mg, 274 mg theoretical, 67%): $R_f 0.32$ (10% CH₃OH–CH₂Cl₂); $[\alpha]^{25}_D$ +7.8 (c 0.18, CHCl₃); ¹H NMR (CDCl₃, 400 MHz) δ 7.40–7.20 (5H, m), 5.75 (1H, d, J = 7.2Hz), 5.30 (2H, br s), 5.26 (1H, d, J = 8.8 Hz), 5.13 (1H, dd, J = 4.0, 8.8 Hz), 4.83 (2H, q, J = 6.8 Hz), 4.77 (1H, dq, J = 6.8, 7.2 Hz), 4.39 (2H, m), 2.23 (3H, s), 2.20 (3H, s), 1.40 (3H, t, J = 6.8 Hz), 0.92 (3H, t)d, J = 6.8 Hz); ¹³C NMR (100 MHz, CDCl₃) δ 169.5, 165.9, 165.2, 163.5, 153.2, 152.6, 133.7, 128.6, 125.7, 125.6, 111.1, 78.7, 73.8, 71.2, 62.0, 54.6, 48.9, 14.5, 14.1, 13.2, 11.9; IR (CHCl₃) v_{max} 3419, 3018, 1778, 1729, 1690, 1612, 1570, 1521, 1443, 1368, 1213, 1071, 928, 756 cm⁻¹; FABHRMS (NBA-CsI) m/e 607.0632 (M⁺ + Cs, C₂₂H₂₆N₄O₆S requires 607.0627).

Ethyl 2-[1(R)-Hydroxy-2.[((4R,5S)-4-methyl-5-phenyl-2-oxazolidinyl)carbonyl]ethyl]-6-amino-5-methylpyrimidine-4-carboxylate (42). A solution of 41 (0.043 mmol, 20 mg) in C₆H₆ (0.7 mL) was treated with Bu₃SnH (0.13 mmol, 38 mg, 3.0 equiv) and warmed at 80 °C for 1 h under N₂. The mixture was allowed to cool to 23 °C, and the solvent was evaporated in vacuo. Flash chromatography (SiO₂, 1 × 3 cm, 5% CH₃OH-CH₂Cl₂) afforded 42 as a white film (16 mg, 17 mg theoretical, 93%): R_f 0.32 (10% CH₃OH-CH₂Cl₂); $[\alpha]^{25}_D$ +37.5 (c 0.13, CHCl₃); ¹H NMR (CDCl₃, 400 MHz) δ 7.40-7.20 (5H, m), 5.91 (2H, br s), 5.70 (1H, d, J = 7.2 Hz), 5.15 (1H, dd, J = 3.3, 8.0 Hz), 4.80 (1H, dq, J = 6.8, 7.2 Hz), 4.41 (2H, q, J = 6.8 Hz), 3.61 (1H, dd, J = 3.6, 16.4 Hz), 3.47 (1H, dd, J = 8.0, 16.4 Hz), 2.19 (3H, s), 1.39 (3H, t, J = 6.8 Hz), 0.90 (1H, d, J = 6.8 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ 170.6, 166.2, 165.8, 162.9, 153.1 (2C), 133.2, 128.6, 128.5, 125.6, 110.6, 79.0, 69.0, 61.7, 54.6, 42.6, 14.3, 14.0, 11.5; IR (CHCl₃) ν_{max} 3420, 3018, 1780, 1726, 1612, 1521, 1424, 1214, 1043, 928 cm⁻¹; FABHRMS (NBA-CsI) m/e 561.0747 (M⁺ + Cs, C₂₁H₂₄N₄O₆ requires 561.0750).

3. (6. Amino-4. (ethoxycarbonyl)-5-methylpyrimidin.2. yl).3(*R*)hydroxypropionamide (43). The agent 42 (0.14 mmol, 57 mg) was treated with an ethanolic solution of NH₃ (16%, 8 mL), and the solution was stirred for 1.5 h at 0 °C. The solvent was evaporated in vacuo. Flash chromatography (SiO₂, 1 × 5 cm, 10% CH₃OH-CH₂Cl₂) afforded 43 as a white film (27 mg, 35 mg theoretical, 77%): R_f 0.12 (10% CH₃-OH-CH₂Cl₂); [α]²⁵_D +35 (c 0.10, CH₃OH); ¹H NMR (CD₃OD, 400 MHz) δ 4.82 (1H, br s), 4.30 (2H, q, J = 7.2 Hz), 2.67 (1H, dd, J =3.6, 14.8 Hz), 2.50 (1H, dd, J = 8.8, 14.8 Hz), 2.04 (3H, s), 1.28 (3H, t, J = 7.2 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ 176.2, 168.2, 167.7, 165.7, 154.3, 71.9, 63.1, 43.5, 14.5, 12.3; IR (CHCl₃) ν_{max} 3423, 3018, 1728, 1676, 1522, 1424, 1213, 1046, 928 cm⁻¹; FABHRMS (NBA-CsI) m/e 401.0224 (M⁺ + Cs, C₁₁H₁₆N₄O₄ requires 401.0226).

3-(6-Amino 4-(ethoxycarbonyl)-5-methylpyrimidin-2-yl)-3(S)azidopropionamide (44). Diethyl azodicarboxylate (0.19 mmol, 33 mg, 1.2 equiv) and a 0.8 M benzene solution of hydrazoic acid⁴³ (0.48 mmol, 0.60 mL, 3 equiv) were added sequentially to a solution of 43 (0.16 mmol, 41 mg) and Ph₃P (0.19 mmol, 51 mg, 1.2 equiv) in dry THF (1.2 mL) at -78 °C. The mixture was stirred for 36 h at 0 °C. The solvent was evaporated in vacuo. Flash chromatography (SiO₂, 1×5 cm, 5% CH₃-OH-CH₂Cl₂) afforded 44 as a white film (38 mg, 45 mg theoretical, 84%): R_{f} 0.45 (10% CH₃OH-CH₂Cl₂); $[\alpha]^{25}$ -71.7 (c 0.12, CHCl₃); ¹H NMR (CD₃OD, 400 MHz) δ 4.65 (1H, dd, J = 5.6, 8.8 Hz), 4.31 (2H, q, J = 7.2 Hz), 2.80 (1H, dd, J = 5.2, 14.8 Hz), 2.66 (1H, dd, J)= 9.6, 14.8 Hz), 2.05 (3H, s), 1.29 (3H, t, J = 7.2 Hz); ¹³C NMR (CD₃OD, 100 MHz) & 175.0, 167.5, 165.8, 165.1, 154.8, 112.0, 63.6, 63.0, 39.7, 14.4, 12.1; IR (CHCl₃) v_{max} 3421, 3018, 2107, 1724, 1522, 1424, 1218, 928 cm⁻¹; FABHRMS (NBA) m/e 294.1313 (M⁺, $C_{11}H_{15}N_7O_3$ requires 294.1315).

3-(6-Amino-4-(ethoxycarbonyl)-5-methylpyrimidin-2-yl)-3(5)-aminopropionamide (45). A solution of 44 (0.09 mmol, 25 mg) in CH₃OH (4 mL) was stirred with 10% Pd-C (6 mg) for 1 h under a H₂ atmosphere. The catalyst was removed by filtration through Celite, and the filtrate was concentrated in vacuo. Flash chromatography (SiO₂, 1 × 3 cm, 20% CH₃OH-CH₂Cl₂) afforded 45 as a film (20 mg, 23 mg theoretical, 88%): R_f 0.10 (20% CH₃OH-CH₂Cl₂); $[\alpha]^{25}_{D}$ -20.0 (c 0.12, CH₃OH); ¹H NMR (CD₃OD, 400 MHz) δ 4.31 (2H, q, J = 6.8 Hz), 4.14 (1H, br s), 2.74 (1H, dd, J = 4.0, 15.2 Hz), 2.49 (1H, dd, J = 8.4, 15.2 Hz), 2.04 (3H, s), 1.28 (3H, t, J = 6.8 Hz); ¹³C NMR (CD₃OD, 100 MHz) δ 176.5, 169.1, 167.8, 165.7, 154.6, 111.0, 63.0, 55.0, 43.1, 14.4, 12.0; IR (neat) ν_{max} 3394, 2984, 1718, 1656, 1579, 1446, 1405, 1236, 1077, 1015 cm⁻¹; FABHRMS (NBA-NaI) m/e 290.1220 (M⁺ + Na, C₁₁H₁₇N₅O₃ requires 290.1229).

3-(6-Amino-4-(ethoxycarbonyl)-5-methylpyrimidin $2 \cdot yl$)-3(S)-[(2-(((*tert*-butyloxy)carbonyl)amino)ethyl)amino]propionamide (46). A solution of 45 (0.02 mmol, 5.0 mg) and N-((*tert*-butyloxy)carbonyl)-2-bromoethylamine (0.04 mmol, 9.0 mg, 2.0 equiv) in DMF (45 μ L) was treated with K₂CO₃ (0.025 mmol, 3.5 mg, 1.2 equiv), and the mixture was stirred under Ar at 25 °C for 2 days. The solvent was evaporated in vacuo. Flash chromatography (SiO₂, 10% CH₃OH-CH₂Cl₂) afforded 45 as a film (4.2 mg, 8.2 mg theoretical, 51%): R_f 0.38 (20% CH₃-OH-CH₂Cl₂); $[\alpha]^{25}_{D}$ -22.0 (c 0.075, CH₃OH): ¹H NMR (CD₃OD, 400 MHz) δ 4.29 (2H, q, J = 6.8 Hz), 3.86 (1H, m), 3.02 (2H, m), 2.45 (2H, m), 2.02 (3H, s), 1.28 (9H, s), 1.26 (3H, t, J = 6.8 Hz); 1³C NMR (CD₃OD, 100 MHz) δ 1.76.4, 168.1, 167.8, 165.9, 158.5, 154.5, 80.2, 63.1, 61.6, 48.0, 41.3, 41.1, 28.9, 14.5, 12.1; IR (neat) ν_{max} 3420, 1718, 1638, 1581, 1439, 1259, 865 cm⁻¹; FABHRMS (NBA-CsI) m/e 543.1340 (M⁺ + Cs, C₁₈H₃₀N₆O₅ requires 543.1332).

3-(6-Amino-4-carboxy-5-methylpyrimidin-2-yl)-3(5)-[(2-(((tertbutyloxy)carbonyl)amino)ethyl)amino]propionamide (47). A solution of 46 (0.01 mmol, 4.2 mg) in THF-CH₃OH-H₂O (3:1:1, 0.3 mL) at 0 °C was treated with aqueous 1 N LiOH (0.015 mmol, $15 \,\mu$ L, 1.5 equiv), and the mixture was stirred for 1.5 h. After evaporation of most of the THF-CH₃OH, the aqueous phase was acidified to pH 4-5 with the addition of aqueous 1.2 N HCl and the solvent was evaporated in vacuo. The residue was charged onto a column of Dowex (1 × 2 cm; acetate form, 50-100 mesh). The column was washed with H₂O, and subsequent elution with 6% HOAc-H₂O afforded 47 as a thin film (3.3 mg, 3.9 mg theoretical, 85%): R_f 0.62 (4:1:1 *i*PrOH-H₂O-HOAc); $[\alpha]^{25}_D$ -8.0 (c 0.20, H₂O); ¹H NMR (CD₃OD, 400 MHz) δ 4.20 (1H, br s), 3.18 (2H, m), 2.87 (2H, m), 2.78 (1H, dd, J = 4.4, 16.0 Hz), 2.67 (1H, dd, J = 8.0, 16.0 Hz), 2.02 (3H, s), 1.31 (9H, s); IR (neat) ν_{max} 3333, 2964, 1676, 1574, 1420, 1169 cm⁻¹; FABHRMS (NBA) *m/e* 383.2051 (M⁺ + H, C₁₆H₂₆N₆O₅ requires 383.2043).

3-(6-Amino-4-carboxy-5-methylpyrimidin-2-yl)-3(5)-[(2-aminoethyl)amino]propionamide ((-)-descarboxamidopyrimidoblamic acld, 4). The compound 47 (0.003 mmol, 1.2 mg) was treated with 3 N HCl-EtOAc (0.3 mL), and the mixture was stirred at 25 °C for 1 h. The solvent was evaporated in vacuo to give pure 4 hydrochloride (0.9 mg, 1.0 mg theoretical, 90%) as a clear hygroscopic solid: R_f 0.38 (4:1:1 *i*PrOH-H₂O-HOAc); [α]²⁵_D-17.8 (c 0.045, H₂O); ¹H NMR (D₂O, 400 MHz) δ 4.66 (1H, dd, J = 5.6, 7.2 Hz), 3.36 (4H, m), 3.10 (1H, dd, J = 5.6, 16.8 Hz), 3.02 (1H, dd, J = 7.2, 16.8 Hz), 2.18 (3H, s); IR (neat) $ν_{max}$ 3491, 3199, 1696, 1627, 1542, 1362, 1203 cm⁻¹; FABHRMS (NBA) m/e 283.1520 (M⁺, C₁₁H₁₈N₆O₃ requires 283.1519).

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